



Diamond in Nanoscale Biosensing

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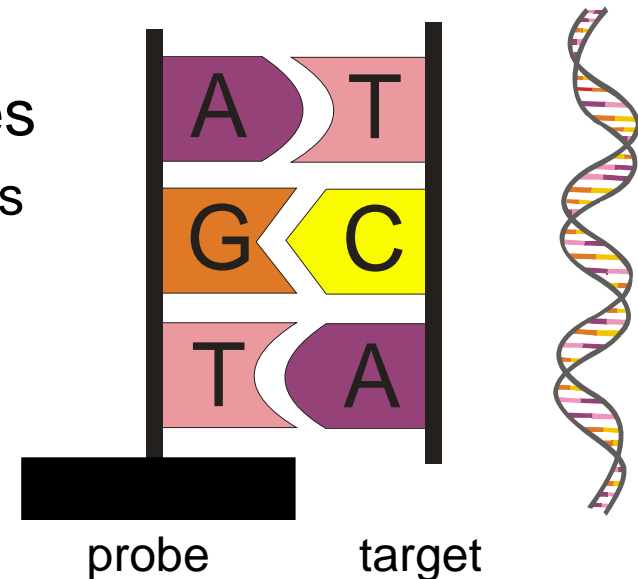
Outline

- Why nanoscale biosensing?
- Why diamond?
 - Hydrogen-terminated and oxidized diamond surfaces
- Attachment of DNA to diamond
 - Photo- and electrochemical methods
 - Fluorescence microscopy
- Structural and mechanical properties of DNA
 - Atomic force microscopy in liquids
 - Optimized detection of DNA thickness (phase shift)
 - Detailed DNA morphology
 - Geometric model → DNA orientation and density
 - Mechanical stability of DNA bonding
- Comparison with other substrate materials
- Conclusion



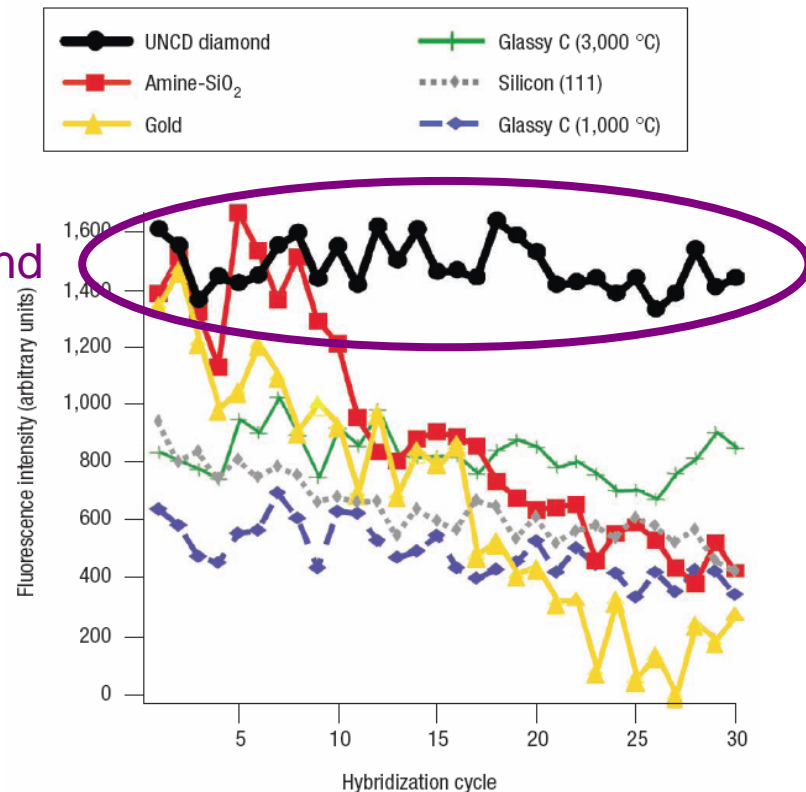
Biosensing

- **Crucial** for health care, medical treatment, drug development, ...
- Typical: recognition of DNA sequences
 - Encoded genetic information and functions
 - Unique matching of base pairs (A-T,C-G)
- Big machines can do it well, but...
- Nanoscale biosensing
 - higher sensitivity ($<fM$), lower cost
 - portable and remote diagnostics (aging society!)
 - Needs substrate to carry DNA
 - Needs new ways of detection



Diamond and biosensing

- Diamond is very interesting for bio-sensors
 - semiconductor (wide band gap)
 - considered highly biocompatible
 - transparent (optical sensing)
 - hard, durable, and stable...
 - well accepted by public



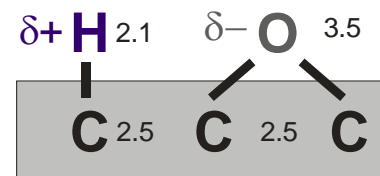
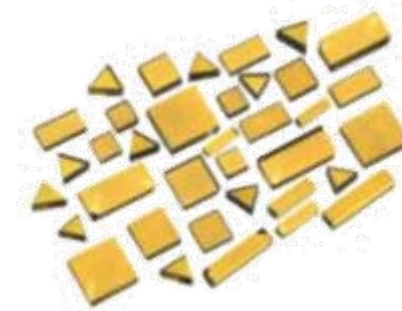
diamond

[W. Yang et al., Nature Mat. 1 (2002) 254]



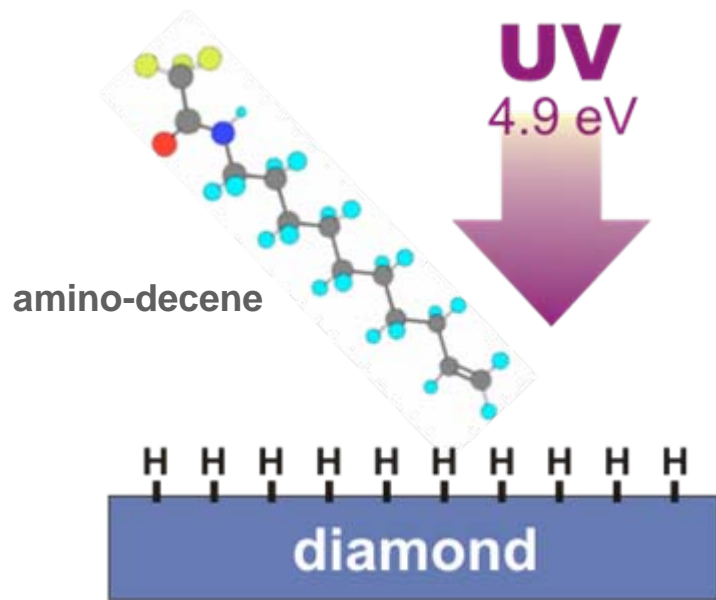
Diamond surface functionalization

- How to make diamond?
 - from methane using plasma assisted chemical vapor deposition (CVD)
 - polycrystalline: on silicon or glass
 - **monocrystalline**: on diamonds (homoepitaxy), advantageous for research
- How to attach molecules to “inert” diamond?
- Surface can be functionalized by atoms
 - plasma techniques, wet chemical techniques
 - we use H-terminated and oxidized surfaces
- Atoms can be replaced by organic molecules
 - Photochemical reactions
 - Electrochemical reactions



Attachment of linker molecules to diamond

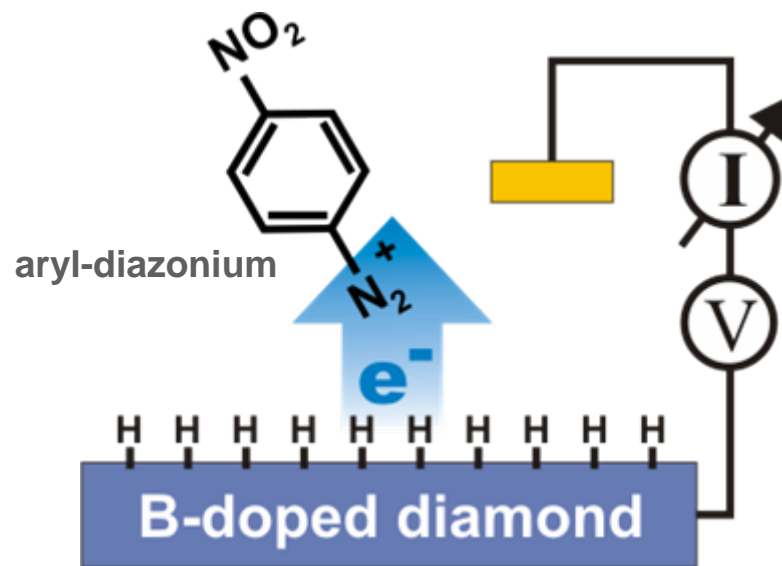
Photochemical



5-7 hours

reaction with hydrogen atoms

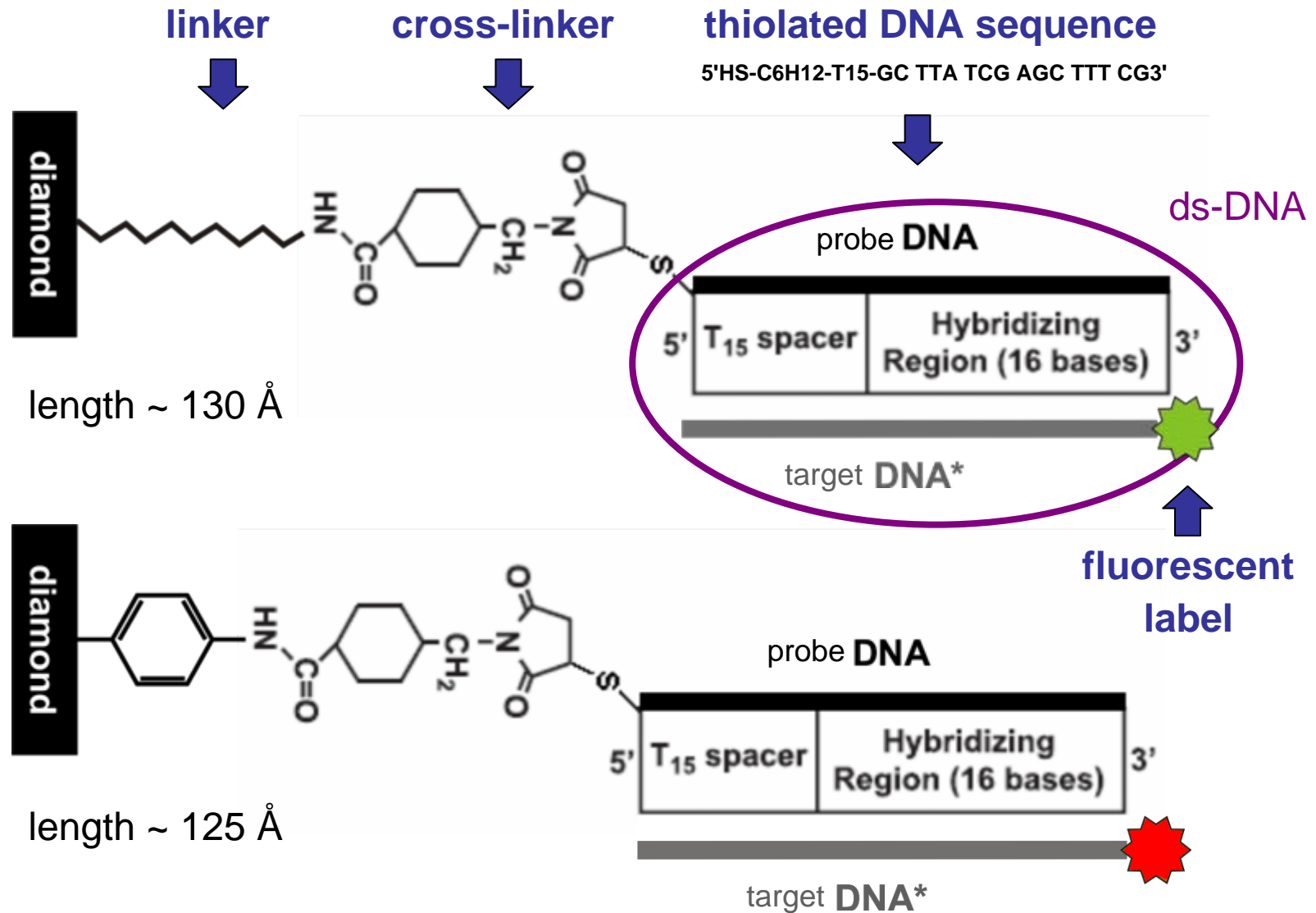
Electrochemical



1 minute

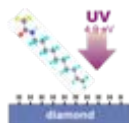
conductive substrate required

Linking of DNA molecules



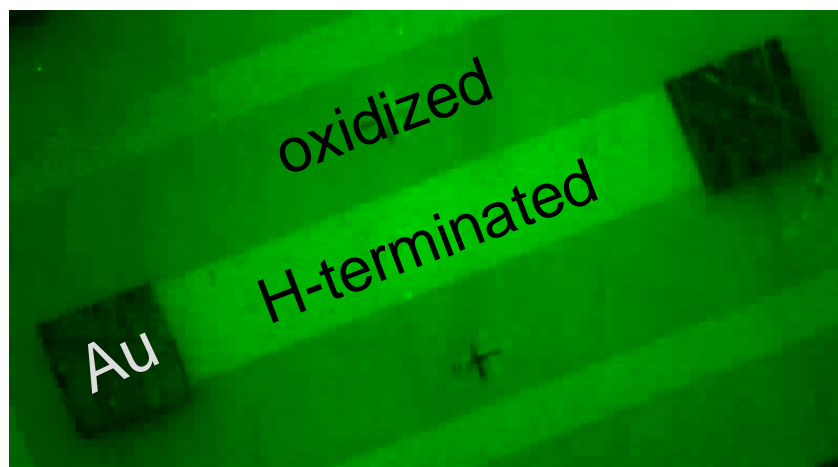
Probing DNA by Fluorescence Microscopy

common technique



Photochemical

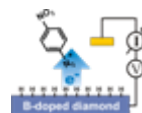
fluorescence (FAM)



100 μm

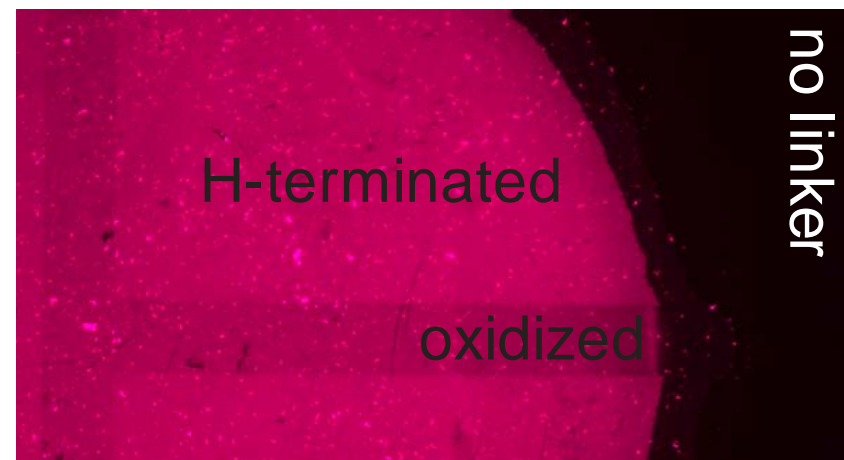
→ DNA present on H-terminated!

(oxidized areas not fully dark)



Electrochemical

fluorescence (Cy5)



100 μm

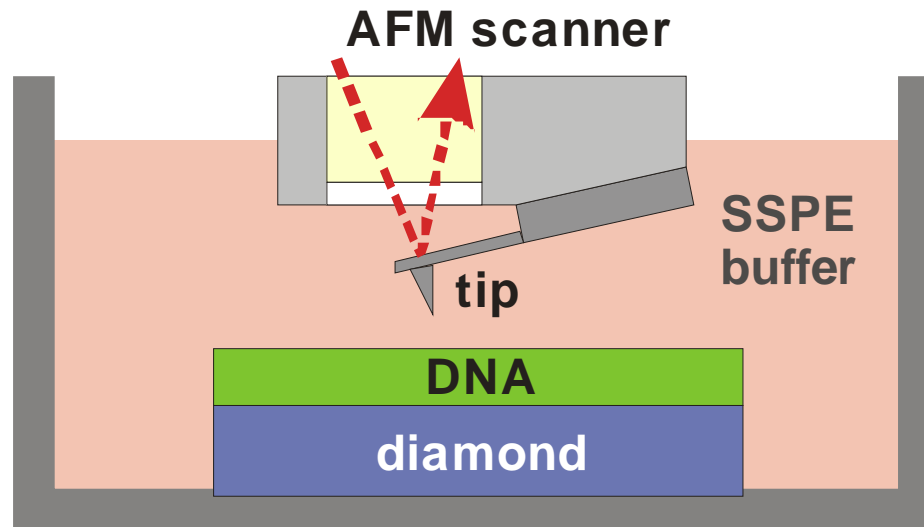
→ DNA present!

Crucial for bio-sensor functionality:
morphology, arrangement, and stability of DNA

beyond abilities
of fluorescence



AFM in buffer solutions



BUFFERS

SSPE/SDS buffer

2x SSPE/ 0.2% SDS buffer, pH=7.4 by NaOH

advantages: bio-environment, no meniscus at tip

AFM TIPS

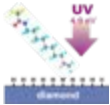
silicon cantilevers (~75kHz in air, ~29kHz in liquid)

force calibration: 56 nN/V

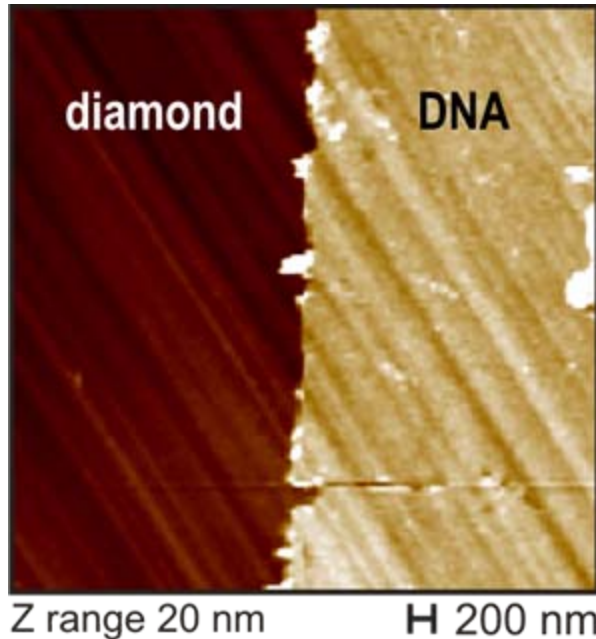
REGIMES

contact (CM-AFM), oscillatory (OM-AFM),
phase detection

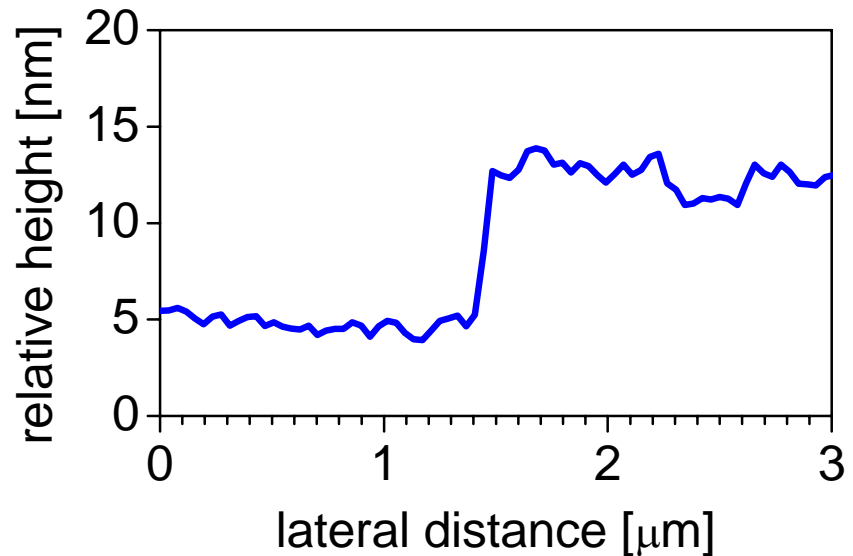
Thickness of DNA layers



OM-AFM image in liquid

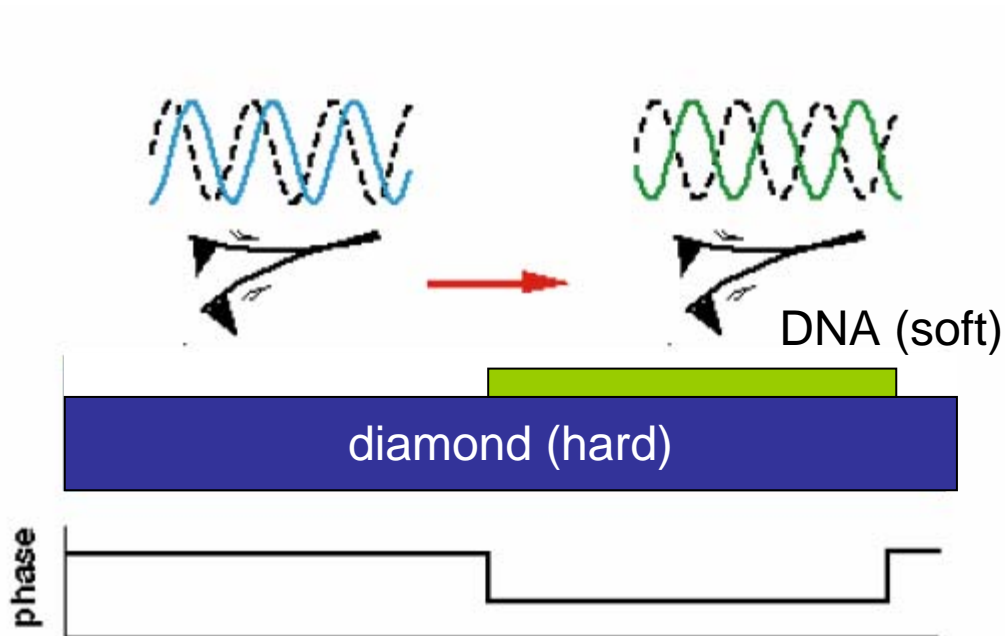


height profile



- step in height $\sim 70\text{-}80 \text{ \AA}$ resolved
- but DNA is soft matter \rightarrow true DNA layer thickness?
- \rightarrow AFM measurement optimized by monitoring phase contrast

Phase contrast in OM-AFM

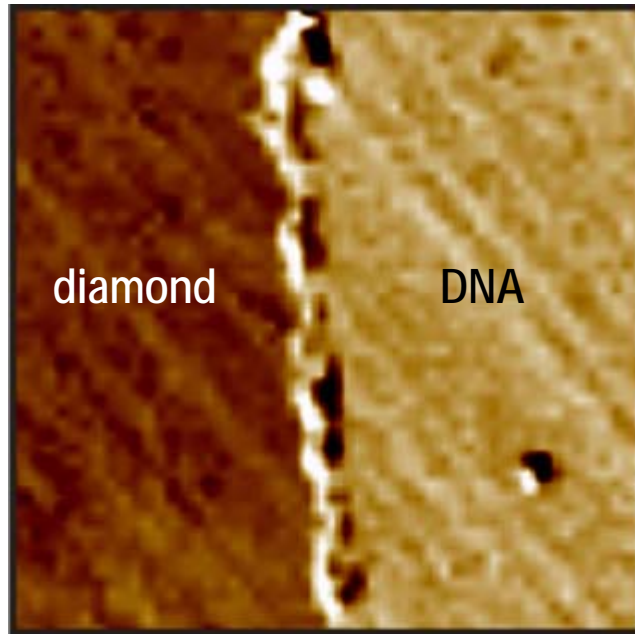


diamond-DNA: phase contrast ~ difference in elastic properties
phase contrast influenced by strength of tip-surface interaction
→ adjusted by **AFM setpoint ratio** (A_{SP}/A_0)

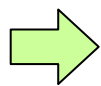
AFM phase images

= 1 for free cantilever

$$A_{SP}/A_0 = 0.40$$

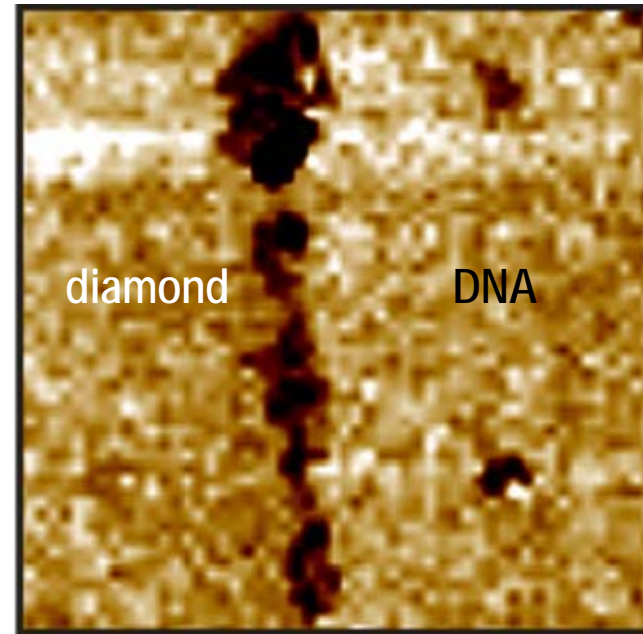


Z range 27° H 100 nm



material contrast
diamond – DNA

$$A_{SP}/A_0 = 0.95$$



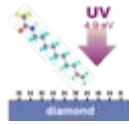
Z range 1.6° H 100 nm



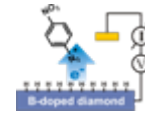
no contrast, DNA
not affected by tip
→ thickness?



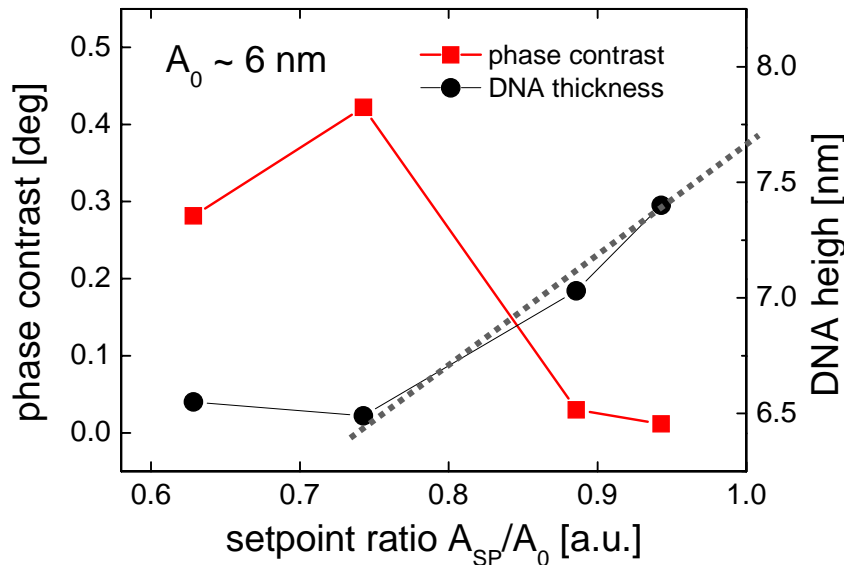
Extrapolated DNA thickness



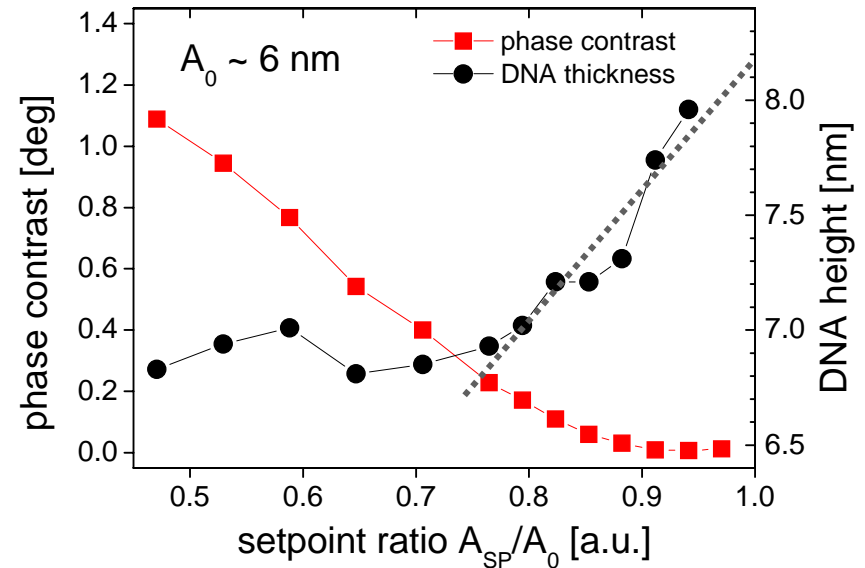
Photochemical



Electrochemical



thickness $\rightarrow (76 \pm 8) \text{ \AA}$



thickness $\rightarrow (82 \pm 5) \text{ \AA}$

As tip-surface distance increases (setpoint ratio $\rightarrow 1$)...

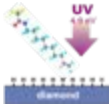
\rightarrow DNA/diamond phase contrast decreases (less tip-DNA interaction)

\rightarrow thickness increases \rightarrow **extrapolated DNA thickness**

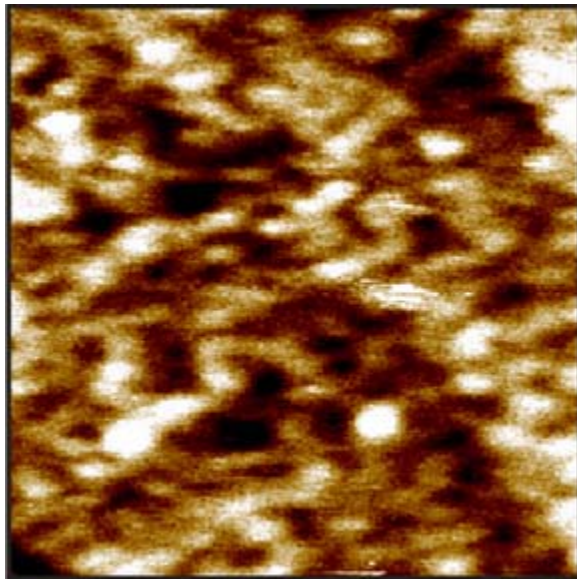
(error bar \sim RMS surface roughness)



DNA layer morphology



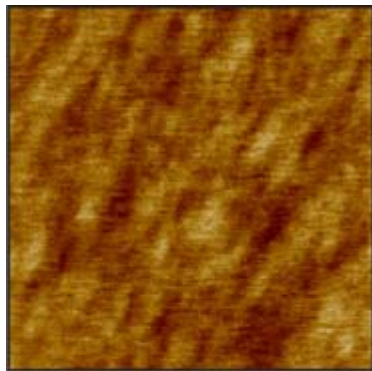
OM-AFM image in liquid



Z range 3 nm

H 10 nm

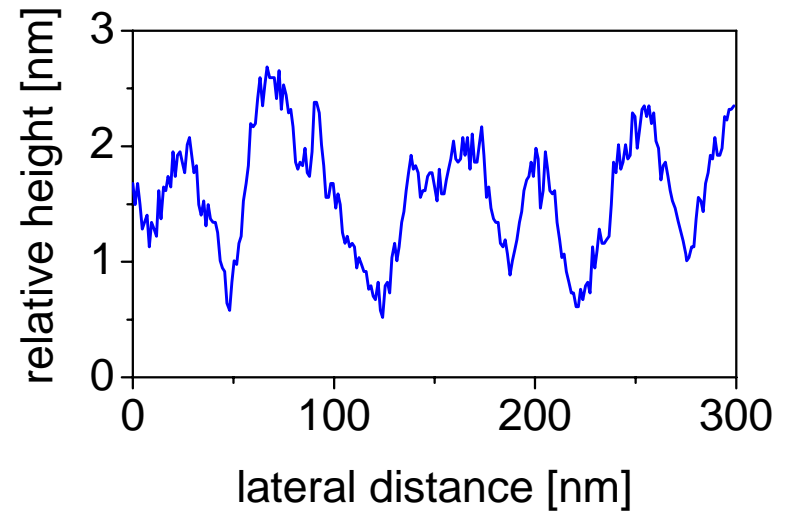
substrate →



Z range 3 nm

H 10 nm

height profile



Features

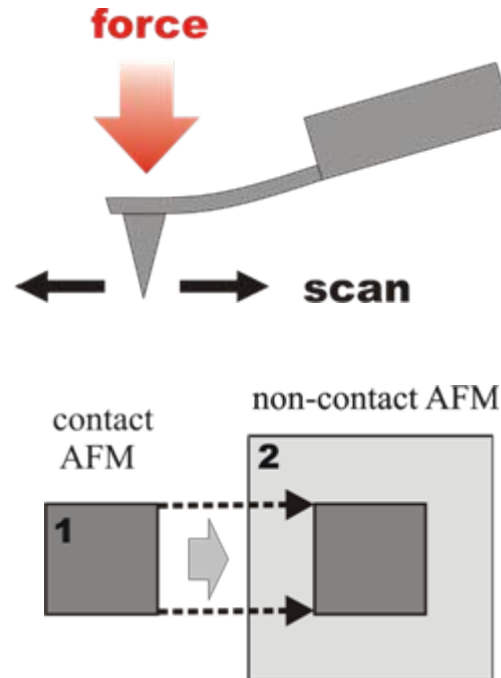
- surface modulations ~ 30 nm width (typical for closely packed DNA)
- roughness RMS ~ 6-8 Å \ll thickness

→ closely packed layer, no pinholes

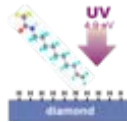
- fine structure ~ DNA?



Mechanical stability of DNA on diamond

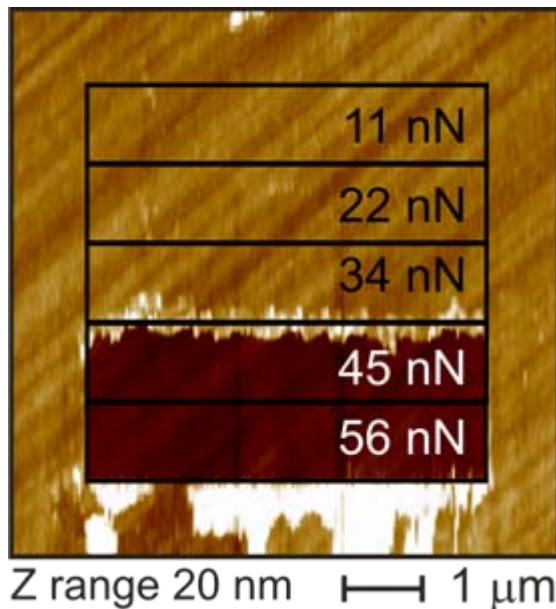


Mechanical stability of DNA on diamond

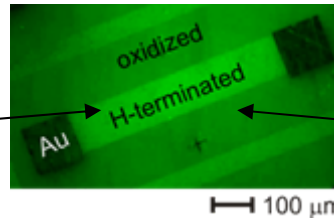


Photochemical

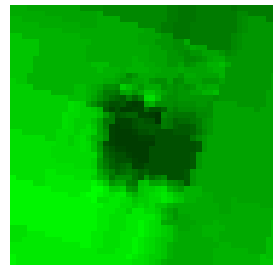
H-terminated



threshold (45 ± 12) nN
→ covalent bonding

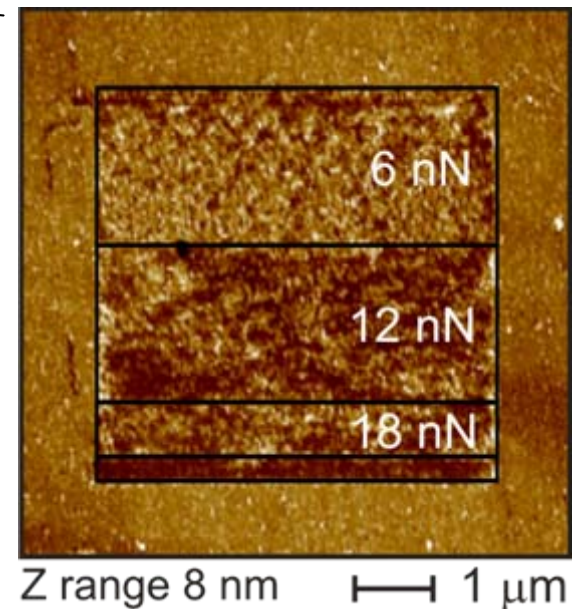


fluorescence:



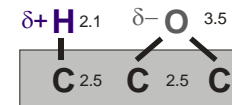
DNA removed

oxidized

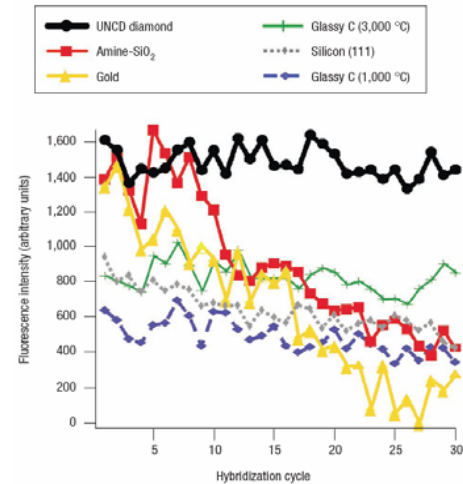
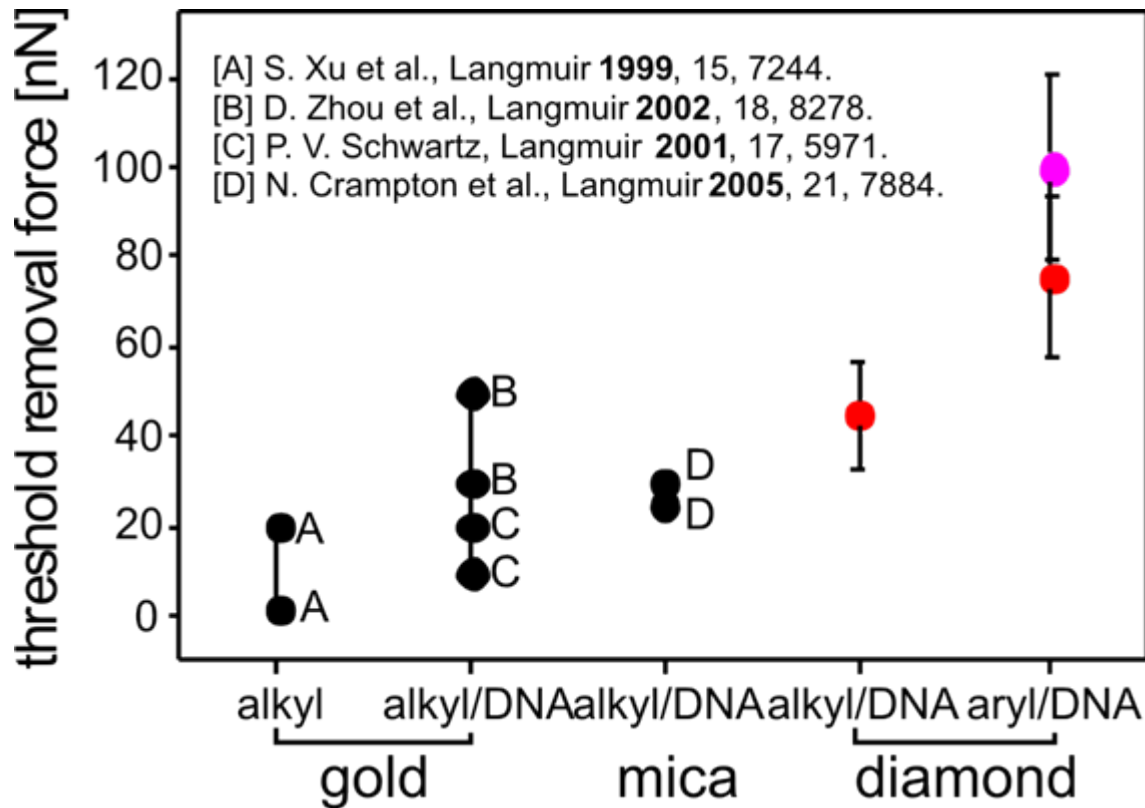


threshold $< (6 \pm 4)$ nN
→ non-covalent bonding (electrostatic?)

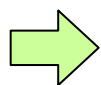
error: interval, spring constant, statistics



Comparison of DNA removal forces



[W. Yang et al., Nature Mat. 1 (2002) 254]



diamond superior with regard to DNA bonding stability, important for bio-sensor reproducibility

note: only approximate comparison, because the parameters and threshold values not clearly specified in the literature



Conclusions

- DNA attached to mono-crystalline diamond
 - by photochemical and electrochemical methods
- Functionality as DNA-sensor demonstrated
 - by fluorescence images of complementary DNA
- Properties of DNA layers on diamond resolved by AFM in liquids:
 - **closely packed**, no pinholes, RMS roughness < 1 nm
 - DNA molecules **inclined**, under angle of 29-36°
 - mechanically **stable** for forces up to 76 nN (good!), indicates **covalent bonding**

	threshold force	DNA thickness	angle	DNA bonding
photochemical process				
H-terminated surface	(45 ± 12) nN	(76 ± 8) Å	31°	covalent
oxidized surface	< (6 ± 4) nN	(20 ± 4) Å	n/a	non-covalent
electrochemical process				
H-terminated surface	(76 ± 18) nN	(82 ± 5) Å	36°	covalent
oxidized surface	(34 ± 9) nN	(69 ± 5) Å	29°	covalent



[B. Rezek et al., *J. Am. Chem. Soc.* **128** (2006) 3884]

