Pseudopotentials for interacting atoms

Extended capabilities of pseudopotential approach based onall-electron pseudopotentials

Contents

- Standard pseudopotentials
 - norm-conserving pseudopotentials
 - ultrasoft pseudopotentials
- Desirable pseudopotential properties
 - "softness"
 - "transferability"
- All-electron pseudopotentials
 - construction process
 - properties, comparison to standard pseudopotentials

What is pseudopotential:

operator simulating the effect (within LDA):
of [NUCLEUS + CORE ELECTRONS]
on electronic states
 in the energy range of interest, e.g. valence states,
 unoccupied states (LDA!) (i.e. not on any state !!!)
 in the real-space region significant for chemical bonds

Requirements, expected properties:

- "sufficient" accuracy in wide energy range (transferability)
- real merit to computational efficiency and/or accuracy
 - by reducing the basis set size (compared to AE)
 - by eliminating large energies of core states

Standard pseudopotentials - history

- Empirical pseudopotentials parameters tuned so that PSP give "good results" for some particular structure; then PSP is assumed to be transferable to another structure
- Ab-initio pseudopotentials
 - 1. free atom calculation; for the atomic state of interest
 - 2. construct pseudowavefunction for a given l and E

$$\psi_{l,E}^{\mathrm{PS}}(r) \equiv \psi_{l,E}^{\mathrm{AE}}(r), \ r > R_C$$

- 3. obtain pseudopotentials by inverted Schrödinger (Dirac) equation
- 4. "unscreen": $V_l^{\text{ion}}(r) = V_l^{\text{scr}}(r) V_{\text{H,XC}}[\rho^{\text{ps}}](r)$.

Ab-initio pseudopotential

Norm-conserving pseudopotentials
 Top and Hopfield [1973], Hamann et al[1979]

$$\frac{\partial}{\partial \epsilon} \frac{\partial}{\partial r} \ln \psi_{\epsilon,l}(r) \Big|_{\epsilon=E,r=R_C} = \frac{-2}{R_C^2 \psi_{E,l}^2(R_C)} \int_0^{R_C} \left| \psi_{E,l}(r) \right|^2 r^2 dr$$



norm-conserving PSP - continued

$$\int_{0}^{R_{C}} \left| \psi_{E,l}^{\mathrm{PS}} \right|^{2} r^{2} dr = \int_{0}^{R_{C}} \left| \psi_{E,l}^{\mathrm{AE}} \right|^{2} r^{2} dr$$

$$\Downarrow$$

$$\frac{\partial}{\partial \epsilon} \frac{\partial \ln \psi_{\epsilon,l}^{\mathrm{PS}}(r)}{\partial r} \bigg|_{\epsilon=E,r=R_{C}} = \frac{\partial}{\partial \epsilon} \frac{\partial \ln \psi_{\epsilon,l}^{\mathrm{AE}}(r)}{\partial r} \bigg|_{\epsilon=E,r=R_{C}}$$

Equal amount of charge of ψ^{AE} and ψ^{PS} implies equal logarithmic derivative, i.e. equal phase shift and equal scattering properties in the neighbourhood of E_{ref} to the 1-st order

Ab-initio pseudopotentials - continued

- ... several various types of norm-conserving PSP Generalized ... [Hamann 1989], [RRKJ 1990], Extended ... [Shirley et al 1990]
- Ultrasoft pseudopotentials Vanderbilt [1990]; relaxing norm-conserving condition \Rightarrow softer PSP, smaller R_C
- PAW (Projector Augmented Wave) method Blöchl [1995]; (-)partial waves in the basis set, (-)still frozen core, (+)unambiguous assignment between AE and PS quantities

Desirable PSP properties ... and problems

1. "softness"

– means the size of basis (as small as possible) set that we need to achieve required accuracy in a given application; is there any independent criterion ? not discoverd so far, but we have some "indicators":

• Visual softness, curvature at r = 0



- Fourier image of PS-potential in reciprocal space
- Fourier image of PS-wave function in reciprocal space

2. "transferability"

intuitive meaning: the ability to work properly in different environments (solids, compounds, molecules), e.g. Na in metal Na and NaCl

two main components:

- (a) the precision of reproducing the scattering properties of AE potential (log. derivative) as a function of
 e in some neighborhood of *E*_{ref} (energy transferability)
- (b) the precision of reproducing the AE eignevalues under varying the external environmental conditions (i.e. charge density within DFT)

transferability ... - continued

(a) and (b) related via the *higher momenta* of the charge density [Shirley et al 1989] (related to higher energy derivatives of the phase-shift):

norm-conserving PSP

- \Rightarrow correct 1-st energy derivative of scattering properties
- ⇒ correct PSP behavior with respect to making *first order* changes to external charge density

 \equiv causing constant potential shift

Sources of errors in pseudopotential method

- Inearized method \Rightarrow higher order errors:
 - energy bands in a solid doesn't coincide with atomic eignevalues

- external charge density (caused by neighboring atoms) is not a *first order* change (doesn't cause constant potential shift)

 \Rightarrow attempts to mimick the chemical bond by choosing suitable atomic configuration for generating pseudopotential

next step in psedopotential construction: subtracting the XC-potential of valence charge density: can cause even first order error since the XC-term is not linear; reduced by NLCC (non-linear core correction)

All electron pseudopotential (AEPP)

 selfconsistent pseudopotential-generating scheme that takes into account the solid state environment:

- 1. all-electron atom recontruction using the crystal boundary conditions
- 2. constructing screened PSP: by minimizing a functional assembled from conditions to be satisfied
- 3. unscreening the PSP by *real solid-state valence charge density*

AEPP - 1.atomic reconstruction

1. crystal charge density forms a *boundary condition*

$$\frac{d}{dr} \ln \left(r^2 \left| \psi_{E_{\text{val},l},l}^{\text{at}}(r) \right|^2 \right) \Big|_{r=R_C} = \frac{d}{dr} \ln \left(r^2 \rho_l^{\text{sps}}(r) \right) \Big|_{r=R_C},$$

where partial charge density in a solid $\rho_l^{\text{sps}}(r)$ is evaluated by summing over all occupied states,

$$\rho_l^{\rm sps}(r) = \sum_{\vec{k},n} \sum_{m=-l}^{l} \frac{1}{4\pi r^2} \int_{SPH} d\Omega d\Omega'$$

$$\psi_{\vec{k},n}^*(r\hat{\mathbf{n}}) Y_{lm}(\hat{\mathbf{n}}) Y_{lm}^*(\hat{\mathbf{n}}') \psi_{\vec{k},n}(r\hat{\mathbf{n}}')$$

The boundary condition (above) replaces the standard condition for the wavefunctions to be normalizable $(\psi_{E,l}(r) \rightarrow 0 \text{ for } r \rightarrow \infty)$ and determines the eigenvalue E_l .

AEPP - 1. atomic reconstruction - continued

The normalization condition for the valence atomic-like radial wavefunctions is

$$\int_{0}^{R_{C}} \left| \psi_{E_{\text{val},l},l}^{\text{at}}(r) \right|^{2} r^{2} dr = \int_{0}^{R_{C}} \rho_{l}^{\text{sps}}(r) r^{2} dr$$

(reversed norm-conserving condition)

The core states are recalculated selfconsistently with the valence (augmented to the crystal charge density) states.

AEPP - 2. screened PSP

2. each component $V_l^{\rm scr}(r)$ satisfies

(i) At $r = R_C$ the potential $V_l^{\text{scr}}(r)$ matches the all-electron potential $V^{\text{at}}(r)$ up to the second derivative,

(ii) At $r = R_C$ the radial pseudo-wavefunctions $R_{E_{n,l},l}^{ps}(r)$

match the corresponding atomic-like radial functions by their values and first derivatives (for each energy window), and

(iii) the correct energy derivative of the

pseudo-wavefunction is ensured by the norm-conserving condition (for main -valence- energy window; relaxed for semicore state)

AEPP - continued

differences with respect to conservative methods:standard methodsAEPP

 boundary conditions for initial atomic calculation: derived from partial free, isolated atom *l*-projected charge density in a solid

 chemical bond is taken into account in PSP by intuitive, ad-hoc choice of atomic occupation numbers
 in natural way via the boundary condition

transferability ranges are located
 around the atomic around centers of l-projected DOS (bands) in a solid

- core states atomic, frozen-core ap- self-consistent with the proximation
 charge density in a solid
- screening XC potential term atomic valence charge self-consistent charge density, NLCC density in a solid