

# Deterministic and stochastic models of dynamics of chemical systems

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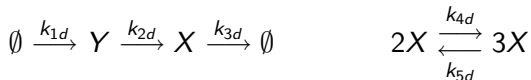
PANM 14, June 1–6 2008, Dolní Maxov

# Outline



- ▶ Chemical system with SNIPER bifurcation
- ▶ Stationary distribution
- ▶ Period of oscillations

# Chemical system with SINIPER bifurcation



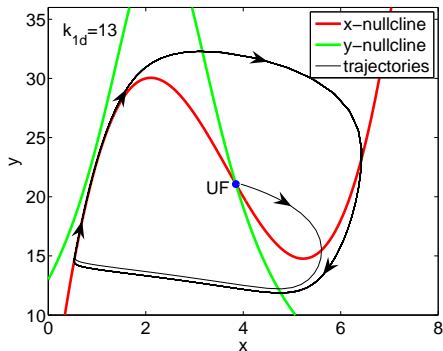
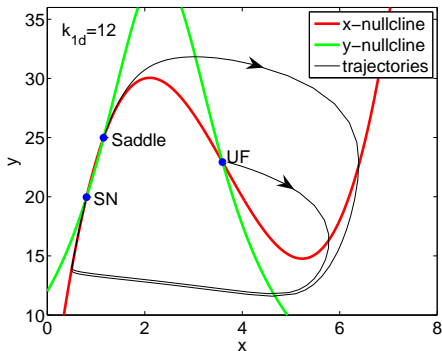
Mean-field ODE:

$$\begin{aligned} \frac{d\tilde{x}}{dt} &= k_{2d}\tilde{y} - k_{5d}\tilde{x}^3 + k_{4d}\tilde{x}^2 - k_{3d}\tilde{x} \\ \frac{d\tilde{y}}{dt} &= -k_{7d}\tilde{x}^2\tilde{y} + k_{6d}\tilde{x}\tilde{y} - k_{2d}\tilde{y} + k_{1d} \end{aligned}$$

$X = X(t)$ ,  $Y = Y(t)$  ... number of molecules

$\tilde{x} = X/V$ ,  $\tilde{y} = Y/V$  ... concentrations  $V$  ... volume

# SNIPER bifurcation



$$k_{1d} = 12 \quad [\text{sec}^{-1} \text{mm}^{-3}]$$

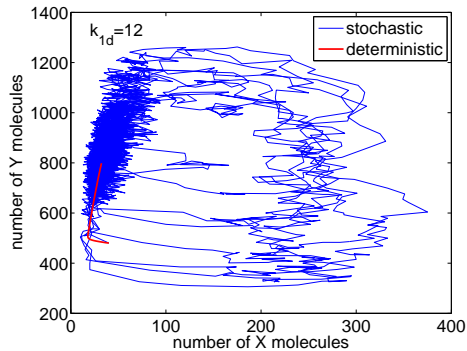
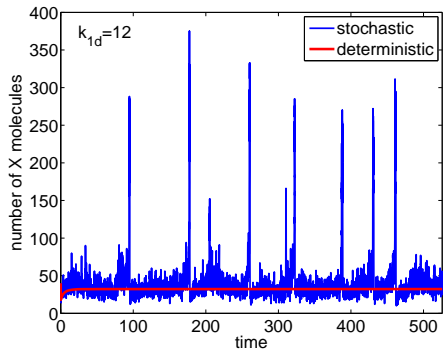
$$k_{2d} = 1 \quad k_{3d} = 33 \quad [\text{sec}^{-1}]$$

$$k_{4d} = 11 \quad k_{6d} = 0.6 \quad [\text{sec}^{-1} \text{mm}^3]$$

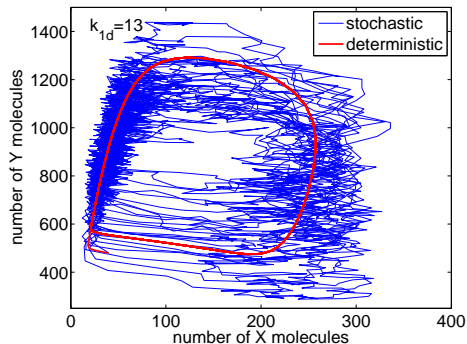
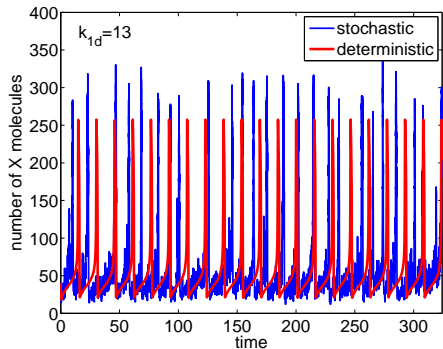
$$k_{5d} = 1 \quad k_{7d} = 0.13 \quad [\text{sec}^{-1} \text{mm}^6]$$

$$k_{1d} = 13 \quad [\text{sec}^{-1} \text{mm}^{-3}]$$

# Stochastic simulations $k_{1d} = 12$



# Stochastic simulations $k_{1d} = 13$



# Propensities and rate constants



$$\alpha_1(x, y) = k_1$$

$$k_1 = k_{1d}V$$

$$\alpha_2(x, y) = k_2y$$

$$k_2 = k_{2d}$$

$$\alpha_3(x, y) = k_3x$$

$$k_3 = k_{3d}$$

$$\alpha_4(x, y) = k_4x(x - 1)$$

$$k_4 = k_{4d}/V$$

$$\alpha_5(x, y) = k_5x(x - 1)(x - 2)$$

$$k_5 = k_{5d}/V^2$$

$$\alpha_6(x, y) = k_6xy$$

$$k_6 = k_{6d}/V$$

$$\alpha_7(x, y) = k_7x(x - 1)y$$

$$k_7 = k_{7d}/V^2$$

$\alpha_i(X(t), Y(t))dt =$  probability that  $i$ -th reaction occurs in  $(t, t + dt)$



## “Naive” stochastic simulations

- for  $i = 1, 2, \dots, 7$  do
  - ▶  $r_i \dots$  random number uniformly distributed in  $(0, 1)$
  - ▶  $p_i = \alpha_i \Delta t$
  - ▶ if  $0 < r_i < p_i \Rightarrow$  Reaction  $i \Rightarrow$  update numbers of reactants and productsend
- $t := t + \Delta t$ , go to 1.

Reaction 1 :  $Y(t + \Delta t) = Y(t) + 1$

Reaction 2 :  $X(t + \Delta t) = X(t) + 1$   $Y(t + \Delta t) = Y(t) - 1$

Reaction 3 :  $X(t + \Delta t) = X(t) - 1$

Reaction 4 :  $X(t + \Delta t) = X(t) + 1$

Reaction 5 :  $X(t + \Delta t) = X(t) - 1$

Reaction 6 :  $Y(t + \Delta t) = Y(t) + 1$

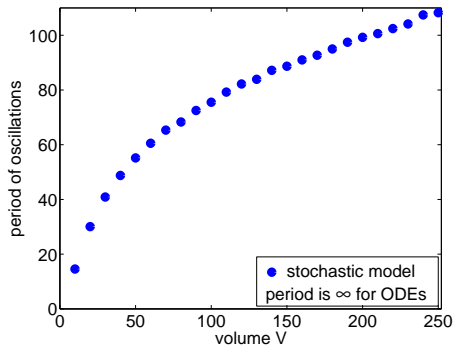
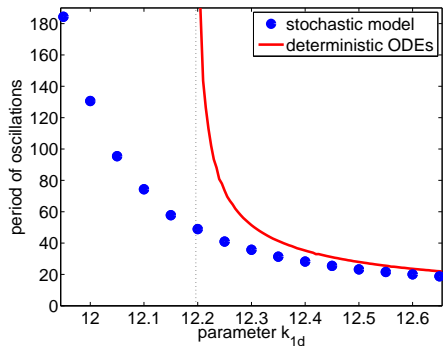
Reaction 7 :  $Y(t + \Delta t) = Y(t) - 1$





1.  $r_1, r_2 \dots$  two random numbers uniformly distributed in  $(0, 1)$
2.  $\alpha_0 = \sum_{i=1}^7 \alpha_i(t) \dots$  where  $\alpha_i(t)$  propensities of all reactions
3.  $\tau = \frac{1}{\alpha_0} \ln \left( \frac{1}{r_1} \right) \dots$  next reaction takes place at  $t + \tau$
4. Find  $j$  such that  $\frac{1}{\alpha_0} \sum_{i=1}^{j-1} \alpha_i(t) \leq r_2 < \frac{1}{\alpha_0} \sum_{i=1}^j \alpha_i(t)$ .  
Reaction  $j \Rightarrow$  update numbers of reactants and products
5.  $t := t + \tau$ , go to 1.

# Period of oscillations



# Chemical Fokker-Planck equation



$$\frac{\partial P}{\partial t} = \frac{\partial^2}{\partial x^2} [d_x P] + \frac{\partial^2}{\partial x \partial y} [d_{xy} P] + \frac{\partial^2}{\partial y^2} [d_y P] - \frac{\partial}{\partial x} [v_x P] - \frac{\partial}{\partial y} [v_y P]$$

$P = P(x, y, t)$  ... probability that  $X(t) = x$  and  $Y(t) = y$

$$v_x(x, y) = \alpha_2(x, y) - \alpha_3(x, y) + \alpha_4(x, y) - \alpha_5(x, y)$$

$$v_y(x, y) = \alpha_1(x, y) - \alpha_2(x, y) + \alpha_6(x, y) - \alpha_7(x, y)$$

$$d_x(x, y) = [\alpha_2(x, y) + \alpha_3(x, y) + \alpha_4(x, y) + \alpha_5(x, y)]/2$$

$$d_y(x, y) = [\alpha_1(x, y) + \alpha_2(x, y) + \alpha_6(x, y) + \alpha_7(x, y)]/2$$

$$d_{xy}(x, y) = -\alpha_2(x, y)$$

# Stationary distribution



$$0 = \frac{\partial^2}{\partial x^2} [d_x P_s] + \frac{\partial^2}{\partial x \partial y} [d_{xy} P_s] + \frac{\partial^2}{\partial y^2} [d_y P_s] - \frac{\partial}{\partial x} [v_x P_s] - \frac{\partial}{\partial y} [v_y P_s]$$

$$\int_0^\infty \int_0^\infty P_s(x, y) dx dy = 1 \quad P_s(x, y) \geq 0 \quad (x, y) \in [0, \infty) \times [0, \infty)$$

$P_s = P_s(x, y) = \lim_{t \rightarrow \infty} P(x, y, t)$  ... stationary distribution

$$-\operatorname{div}(\mathcal{A} \nabla P_s + P_s \mathbf{b}) = 0 \quad \text{in } S = (0, 500) \times (0, 2000)$$

$$(\mathcal{A} \nabla P_s + P_s \mathbf{b}) \cdot \mathbf{n} = 0 \quad \text{on } \partial S$$

$$\mathcal{A} = - \begin{pmatrix} d_x & d_{xy}/2 \\ d_{xy}/2 & d_y \end{pmatrix} \quad \mathbf{b} = \left( v_x - \frac{\partial d_x}{\partial x} - \frac{1}{2} \frac{\partial d_{xy}}{\partial y}, v_y - \frac{\partial d_y}{\partial y} - \frac{1}{2} \frac{\partial d_{xy}}{\partial x} \right)$$

# Stationary distribution



$$0 = \frac{\partial^2}{\partial x^2} [d_x P_s] + \frac{\partial^2}{\partial x \partial y} [d_{xy} P_s] + \frac{\partial^2}{\partial y^2} [d_y P_s] - \frac{\partial}{\partial x} [v_x P_s] - \frac{\partial}{\partial y} [v_y P_s]$$

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$P_s = P_s(x, y) = \lim_{t \rightarrow \infty} P(x, y, t) \dots$  stationary distribution

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$$(\mathcal{A} \nabla P_s + P_s \mathbf{b}) \cdot \mathbf{n} = 0 \quad \text{on } \partial S$$

Remark:

$$-\operatorname{div}(\mathcal{A} \nabla P_s) - \mathbf{b} \cdot \nabla P_s - \operatorname{div}(\mathbf{b}) P_s = 0 \quad \text{in } S$$

$$(\mathcal{A} \nabla P_s) \cdot \mathbf{n} = 0 \quad \text{on } \partial S$$



Classical form

$$\begin{aligned} -\operatorname{div}(\mathcal{A}\nabla P_s + P_s \mathbf{b}) &= 0 \quad \text{in } S = (0, 500) \times (0, 2000) \\ (\mathcal{A}\nabla P_s + P_s \mathbf{b}) \cdot \mathbf{n} &= 0 \quad \text{on } \partial S \end{aligned}$$

Weak form

$$P_s \in H^1(S) : \quad a(P_s, \varphi) = 0 \quad \forall \varphi \in H^1(S)$$

$$a(P_s, \varphi) = \int_S (\mathcal{A}\nabla P_s + P_s \mathbf{b}) \cdot \nabla \varphi \, dx dy$$

FEM

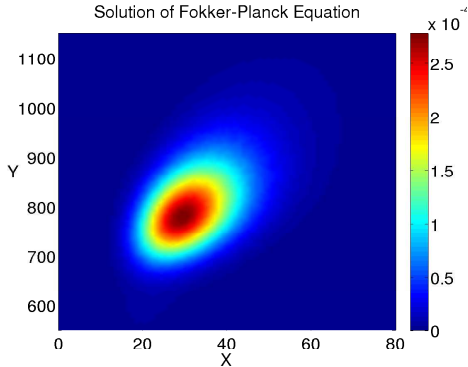
$$P_{s,h} \in W_h : \quad a(P_{s,h}, \varphi_h) = 0 \quad \forall \varphi_h \in W_h$$

$$W_h = \{\varphi_h \in H^1(S) : \varphi_h|_K \in P^1(K), K \in \mathcal{T}_h\}$$

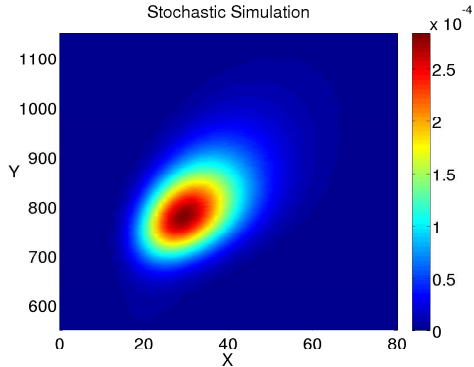
# Stationary distribution – results



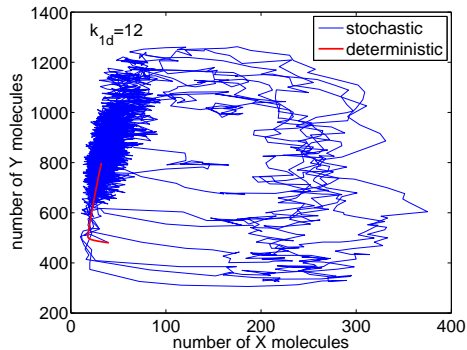
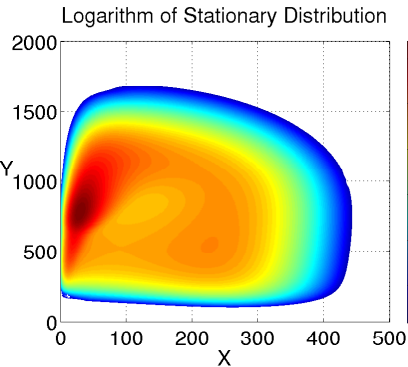
Solution of Fokker-Planck Equation



Stochastic Simulation



# Stationary distribution – results







$$d_x \frac{\partial^2 \tau}{\partial x^2} + d_{xy} \frac{\partial^2 \tau}{\partial x \partial y} + d_y \frac{\partial^2 \tau}{\partial y^2} + v_x \frac{\partial \tau}{\partial x} + v_y \frac{\partial \tau}{\partial y} = -1 \quad \text{for } (x, y) \in \Omega$$

$\tau = \tau(x, y)$  ... average time to leave  $\Omega = \{(x, y) \mid x < 200\}$

Classical form

$$-\operatorname{div}(\mathcal{A} \nabla \tau) + \mathbf{b} \cdot \nabla \tau = -1 \quad \text{in } \tilde{S} = (0, 200) \times (0, 2000)$$

$$\tau = 0 \quad \text{on line } x = 200$$

$$(\mathcal{A} \nabla \tau) \cdot \mathbf{n} = 0 \quad \text{on lines } y = 0, y = 2000, x = 0$$

Weak form

$$\tau \in W : \quad \tilde{a}(\tau, \varphi) = \int_{\tilde{S}} -1 \cdot \varphi \, dx dy, \quad \forall \varphi \in W,$$

$$W = \{v \in H^1(\tilde{S}) : v = 0 \text{ on the line } x = 200\}$$

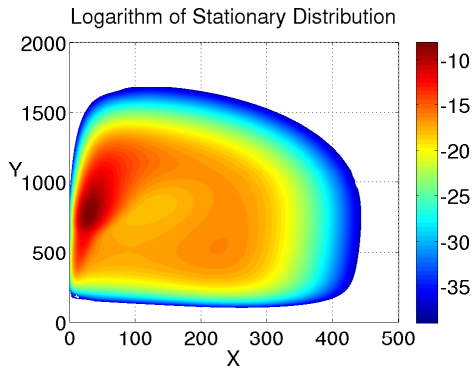
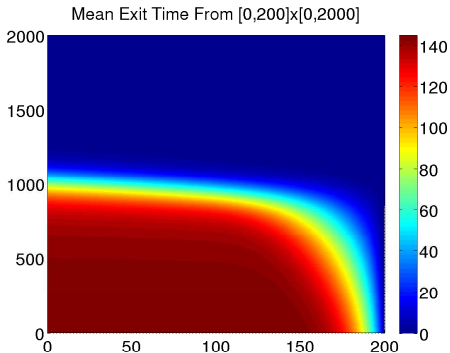
$$\tilde{a}(\tau, \varphi) = \int_{\tilde{S}} \mathcal{A} \nabla \tau \cdot \nabla \varphi \, dx dy + \int_{\tilde{S}} \mathbf{b} \cdot \nabla \tau \varphi \, dx dy.$$

FEM

$$\tau_h \in \tilde{W}_h : \quad \tilde{a}(\tau_h, \varphi_h) = \int_{\tilde{S}} -1 \cdot \varphi_h \, dx dy, \quad \forall \varphi_h \in \tilde{W}_h,$$

$$\tilde{W}_h = \{\varphi_h \in W : \varphi_h|_K \in P^1(K), K \in \mathcal{T}_h\}$$

# Period of oscillations: $k_{1d} = 12$

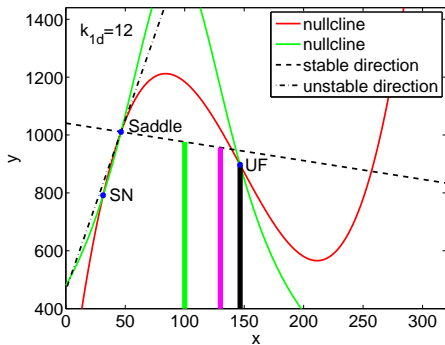
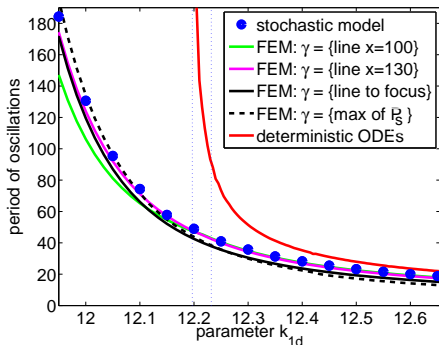


$$\tau_h(29.30, 781.25) \doteq 134.1$$

$$\tau_{stoch,10^5} \doteq 130.4$$

$$T(\gamma) = \frac{\int_{\gamma} \tau(x, y) P_s(x, y) d\gamma}{\int_{\gamma} P_s(x, y) d\gamma}$$

# Period of oscillations: $k_{1d}$ vary

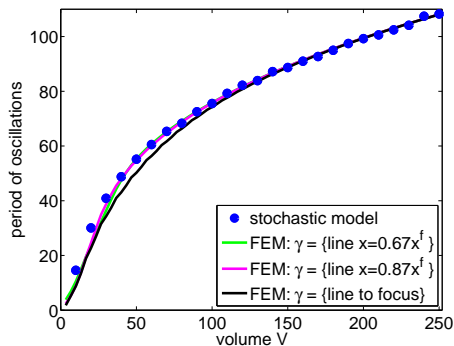


$$\tau_h(29.30, 781.25) \doteq 134.1$$

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$$T(\gamma) = \frac{\int_{\gamma} \tau(x, y) P_s(x, y) d\gamma}{\int_{\gamma} P_s(x, y) d\gamma}$$

Period of oscillations:  $k_{1d} \doteq 12.2$ ; volume vary



$$\tau_h(29.30, 781.25) \doteq 134.1$$

$$\tau_{stoch, 10^5} \doteq 130.4$$

$$T(\gamma) = \frac{\int_{\gamma} \tau(x, y) P_s(x, y) d\gamma}{\int_{\gamma} P_s(x, y) d\gamma}$$

# Conclusions



- ▶ cell cycle modelling
- ▶ more chemical species  $\Rightarrow$  higher dimension

# Thank you for your attention

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