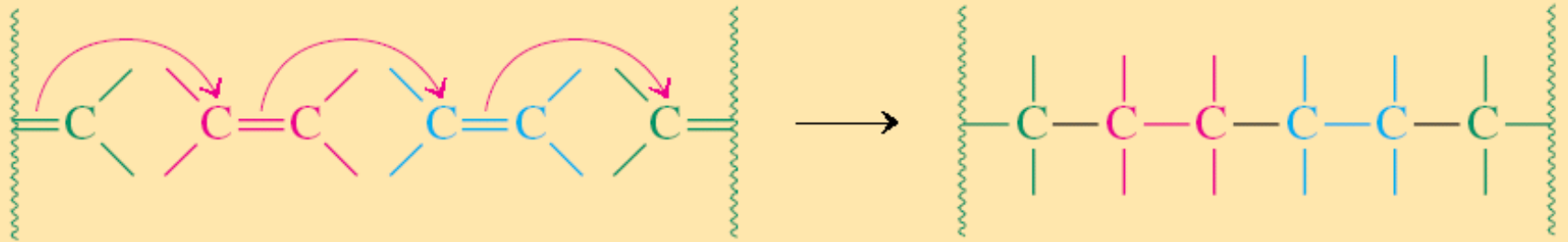


# Polymerizace a polymery



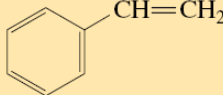


## Polymerization



Monomers

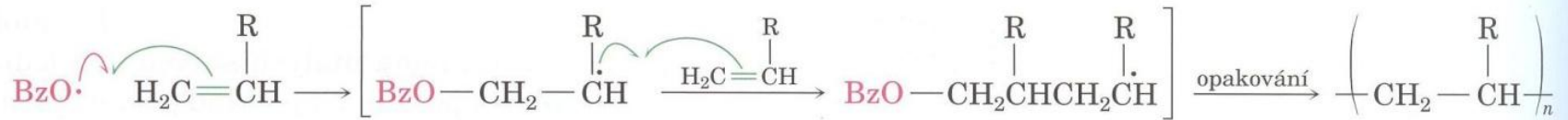
Polymer

**TABLE 12-3 Common Polymers and Their Monomers**

Monomer	Structure	Polymer (common name)	Structure	Uses
Ethene	$\text{H}_2\text{C}=\text{CH}_2$	Polyethylene	$-(\text{CH}_2\text{CH}_2)_n-$	Food storage bags, containers
Chloroethene (vinyl chloride)	$\text{H}_2\text{C}=\text{CHCl}$	Poly(vinyl chloride) (PVC)	$-(\text{CH}_2\underset{\text{Cl}}{\text{CH}})_n-$	Pipes, vinyl fabrics
Tetrafluoroethene	$\text{F}_2\text{C}=\text{CF}_2$	Teflon	$-(\text{CF}_2\text{CF}_2)_n-$	Nonstick cookware
Ethenylbenzene (styrene)		Polystyrene	$-(\text{CH}_2\underset{\text{C}_6\text{H}_5}{\text{CH}})_n-$ 	Foam packing material
Propenenitrile (acrylonitrile)	$\text{H}_2\text{C}=\underset{\text{C}\equiv\text{N}}{\overset{\text{H}}{\text{C}}}$	Orlon	$-(\text{CH}_2\underset{\text{CN}}{\text{CH}})_n-$	Clothing, synthetic fabrics
Methyl 2-methyl- propenoate (methyl methacrylate)	$\text{H}_2\text{C}=\underset{\text{COCH}_3}{\overset{\text{CH}_3}{\text{C}}}$ 	Plexiglass	$-(\text{CH}_2\underset{\text{CO}_2\text{CH}_3}{\overset{\text{CH}_3}{\text{C}}})_n-$	Impact-resistant paneling
2-Methylpropene (isobutylene)	$\text{H}_2\text{C}=\underset{\text{CH}_3}{\overset{\text{CH}_3}{\text{C}}}$	Elastol	$-(\text{CH}_2\underset{\text{CH}_3}{\overset{\text{CH}_3}{\text{C}}})_n-$	Oil-spill clean-up

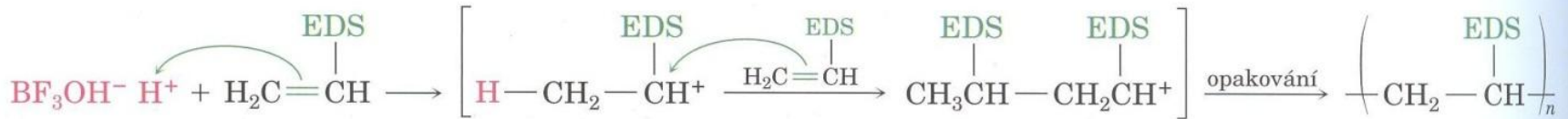
# Polymery s řetězovým růstem

## Radikálová polymerace



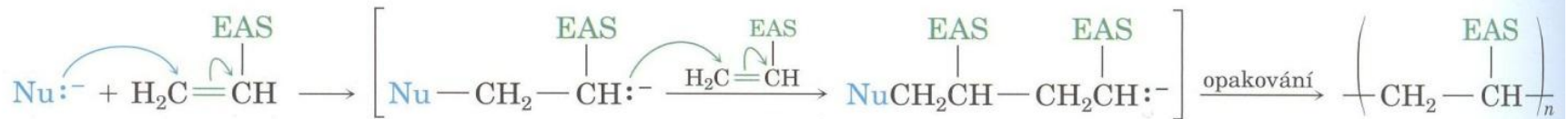
kde  $\text{BzO}\cdot$  = benzoyloxy,  $\text{PhCOO}\cdot$

## Kationtová polymerace



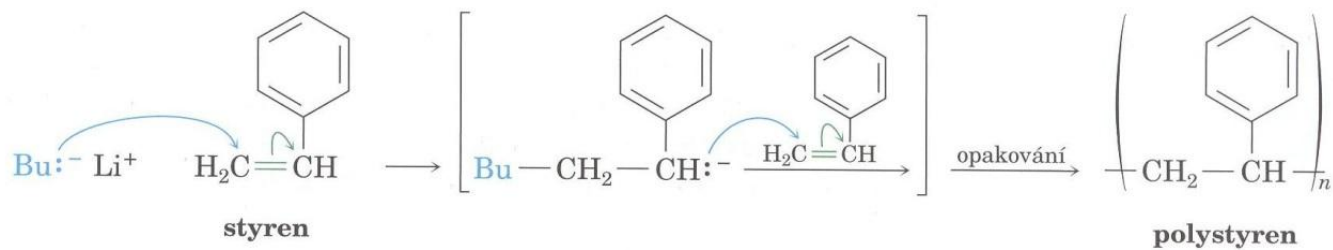
kde  $\text{EDS}$  = elektrondonorní skupina

## Aniontová polymerace

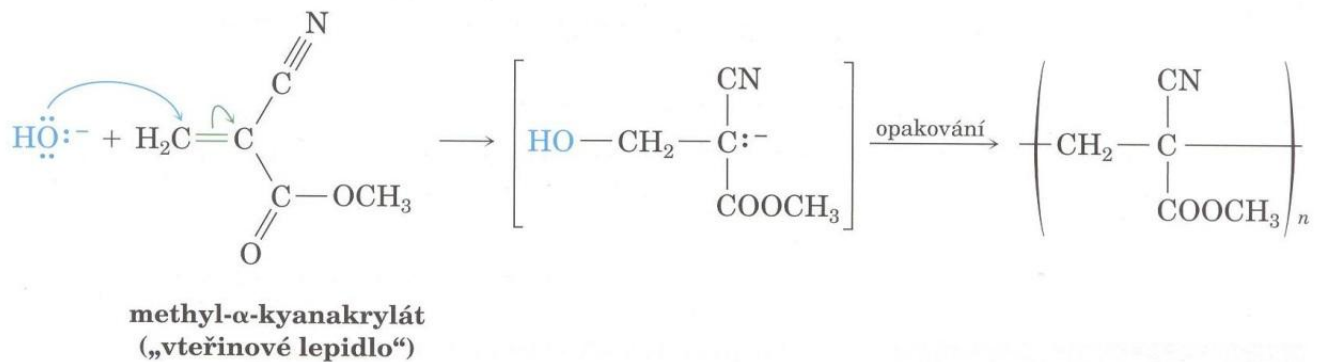


kde  $\text{EAS}$  = elektronakceptorní skupina

## Polystyren

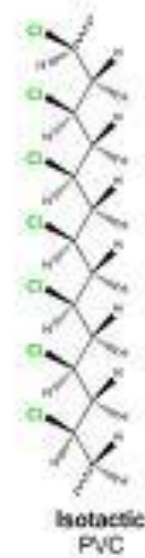
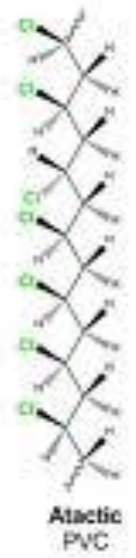
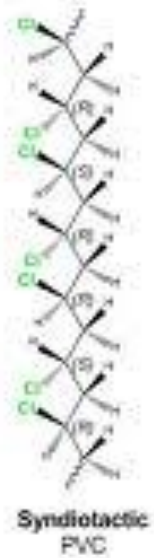
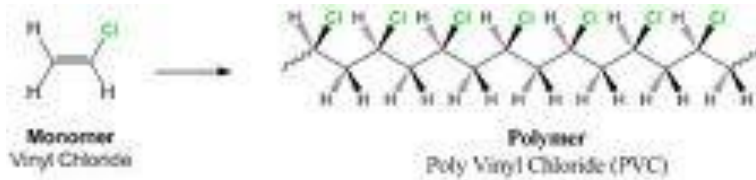
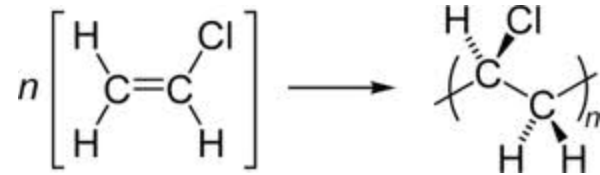


## Polyakryláty – vteřinové lepidlo



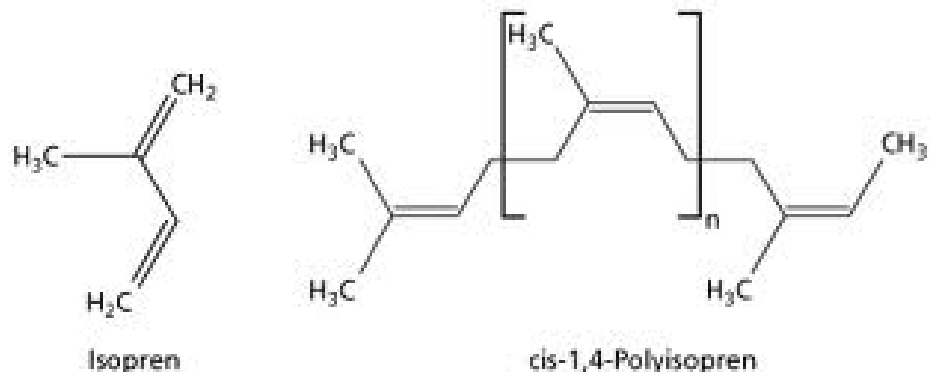


# Polyvinylchlorid - PVC

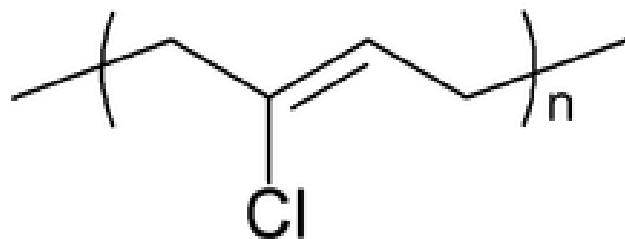




## Polyisopren - kaučuk



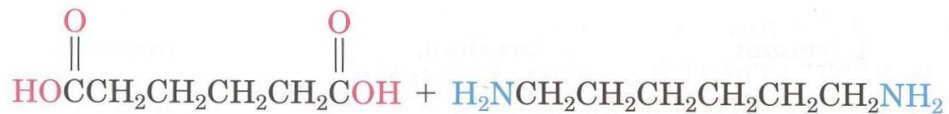
## Polychloropren - neopren





# Polymery se stupňovitým růstem

## Polyamidy



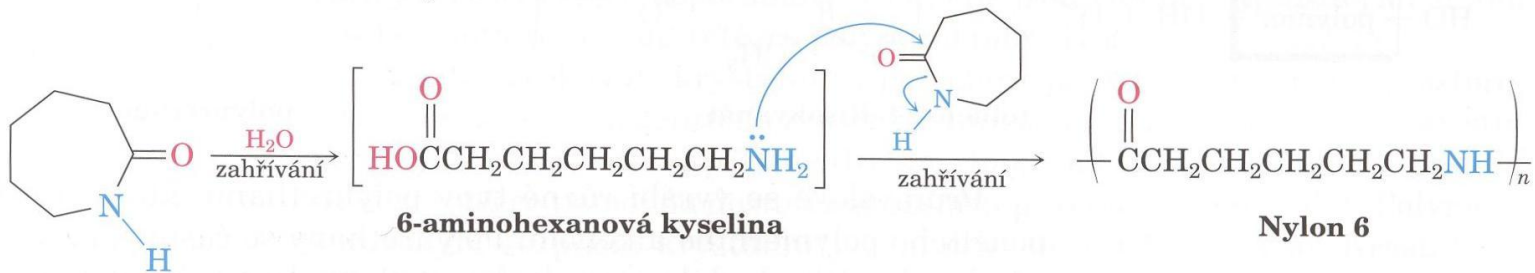
adipová kyselina  
(hexandiová kyselina)

hexamethyldiamin  
(hexan-1,6-diamin)

↓ zahřívání



Nylon 66



6-aminohexanová kyselina

Nylon 6

kaprolaktam

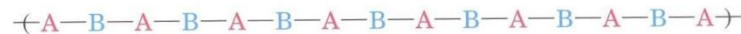
# Kopolymerace a kopolymery

**TABULKA 31.1** Některé běžné kopolymery a jejich použití

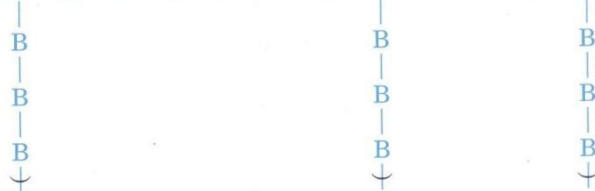
Monomery	Struktura	Název polymeru	Použití
vinylchlorid (20 %) vinylidenchlorid (80 %)	$H_2C=CHCl$ $H_2C=CCl_2$	Saran	obaly na potraviny vlákna
styren (25 %) buta-1,3-dien (75 %)	$H_2C=CHC_6H_5$ $H_2C=CHCH=CH_2$	butadien-styrenový kaučuk	pneumatiky, gumárenské zboží
hexafluorpropen vinylidenfluorid	$F_2C=CF_2$ $H_2C=CF_2$	Viton	těsnění, tmely
akrylonitril buta-1,3-dien	$H_2C=CHCN$ $H_2C=CHCH=CH_2$	nitrilový kaučuk	lepidla, hadice na benzín
isobutylen isopren	$H_2C=C(CH_3)_2$ $H_2C=C(CH_3)CH=CH_2$	butylkaučuk	duše do pneumatik
akrylonitril buta-1,3-dien styren	$H_2C=CHCN$ $H_2C=CHCH=CH_2$ $H_2C=CHC_6H_5$	ABS (iniciály monomerů)	potrubí, nárazuvzdorné materiály



statistický kopolymer

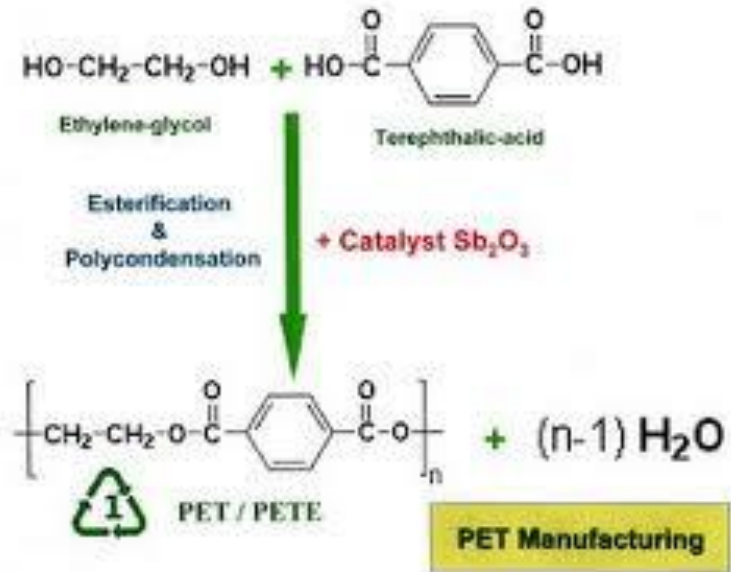


alternující kopolymer

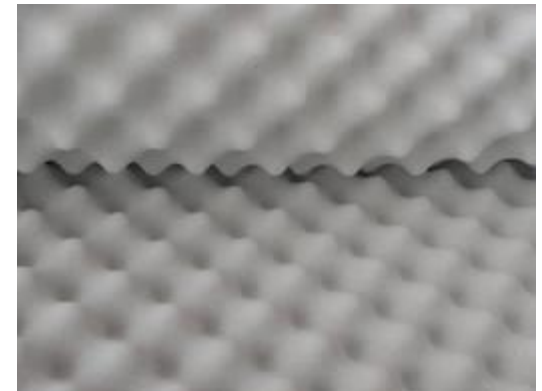
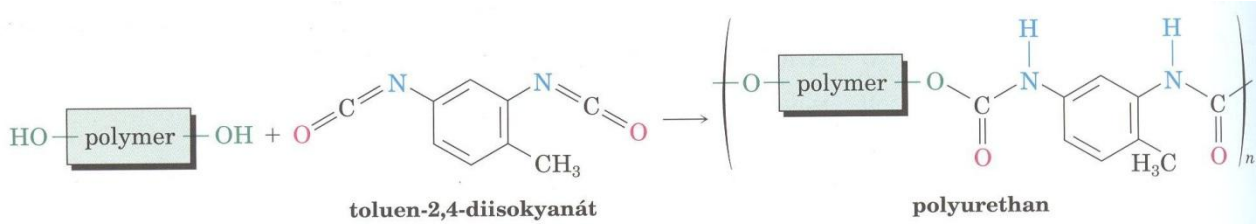
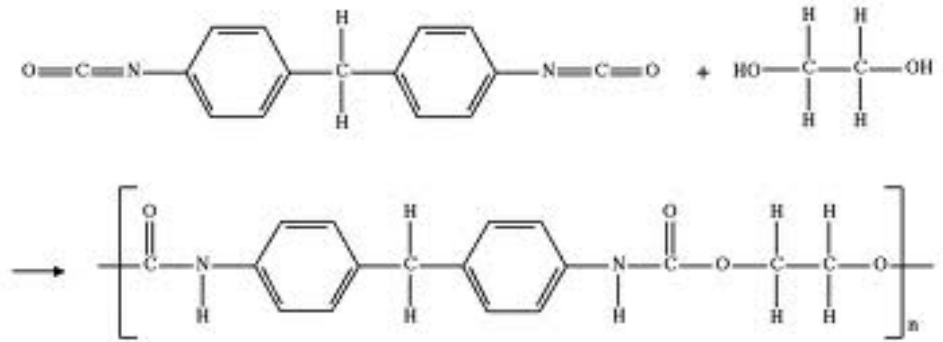
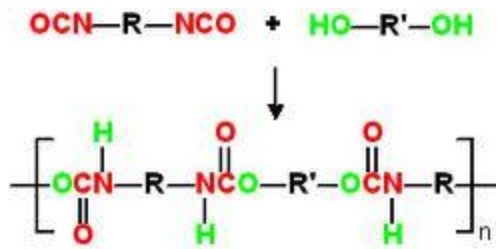


# Polyestery

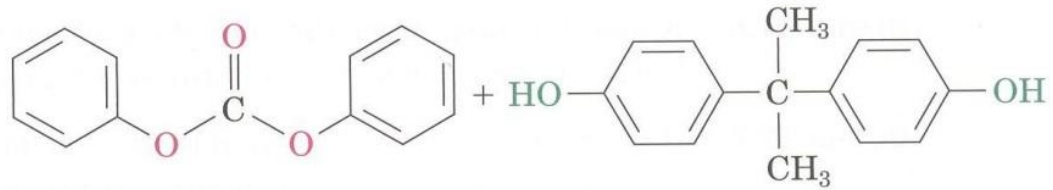
## Polyethylenglykol-tereftalát (PET)



# Polyurethany

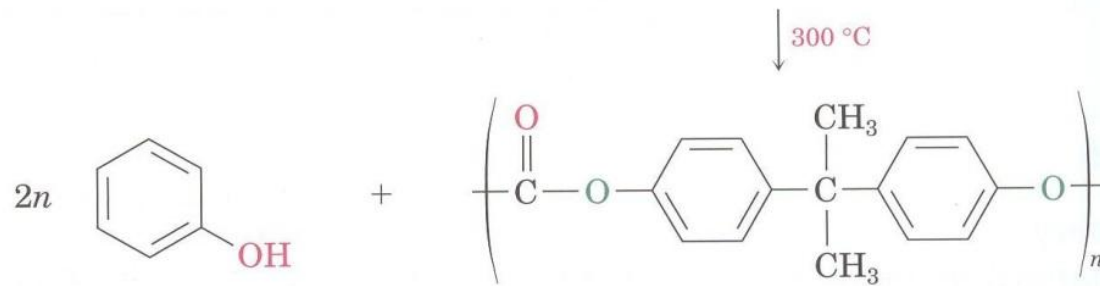


# Polykarbonát



difenyl-karbonát

bisfenol A



Lexan





# Fenol-formaldehydové pryskyřice - bakelit

