The Electrodeless Discharge Lamps Coated with the Titania Thin Film for Photocatalysis in a Microwave Field



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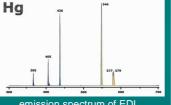
Introduction

Microwave irradiation has been used to assist photochemical and photocatalytic reactions for degradation of organic pollutants. The electrodeless discharge lamp (EDL) generates UV/VIS radiation when placed into the microwave field and is connected to the objective of microwave photochemistry [1]. The degradation of mono-chloroacetic acid (MCAA) by microwave assisted photocatalysis using TiO₂ thin films has been examined. Several factors influencing the degradation of MCAA have been studied in detail [2].

PREPARATION of EDL

The EDL blank was thoroughly cleaned in water-soap mixture, distilled water, aqueous hydrofluoric acid and ethanol. 2.5 µl Hg and stainless steel thin wire were placed to the EDL blank. The system was flushed with argon and sealed under 20 Torr vacuum.



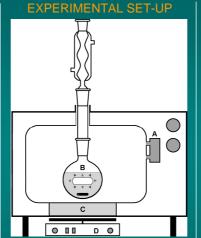


emission spectrum of EDL

$$e^{-} \xrightarrow{MW} e^{-} (accelerated)$$

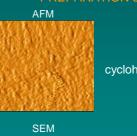
 $e^{-} + Ar \rightarrow Ar^{+} + 2e^{-}, e^{-} + Ar^{+} \rightarrow Ar^{*}$
 $Ar^{*} + Hg \rightarrow Hg^{*} + Ar, Hg^{*} \rightarrow Hg + hv$

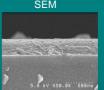
Experimental



batch reactor

A-magnetron B-reaction mixture, EDL C-aluminium plate D-magnetic stirrer





TiO₂

layer

surfactant Triton X-100

neinogenic

cyclohexane water





raw gels, polymeric structure, -O-Ti-O- chains

titanium

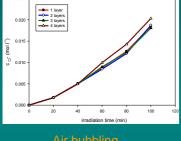
isopropoxide

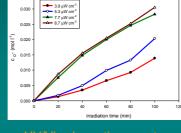
dip-coating method

drying, calcination (400 ℃)

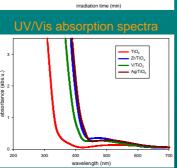
Results

Number of coating cycles

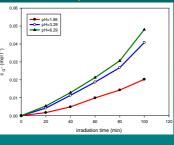




Intensity of light

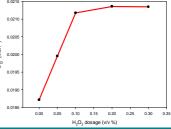


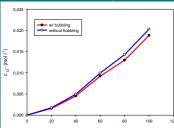
Initial pH value

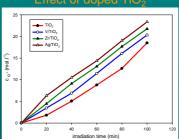


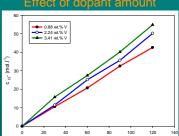


H₂O₂ dosage









Conclusion

The decomposition of MCAA in a MW field was enhanced in an alkaline solution and in the presence of H₂O₂. The UV/VIS spectra of V, Zr and Aq doped TiO₂ show significant absorption in VIS region and the degradation efficiency was higher than those of pure TiO₂.

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The authors thank the Grant agency of the Czech republic for funding this research (Grant No. 104/06/0992). In addition, Hana Žabová thanks the Grant Agency of the Czech Republic for part funding her PhD studentship (Doctoral Grant No. 203/08/H032).

References

[1] P. Klán, V. Církva, Microwaves in Photochemistry, in:A.Loupy, Microwaves in Organic Synthesis, Wiley/VCH, Weinheim, 2006, pp. 860-897. [2] V. Církva, H. Žabová, M. Hájek, J. Photochem. Photobiol. A: Chem. 198 (2008) 13-17.