

Deterministic and stochastic models of circadian rhythms

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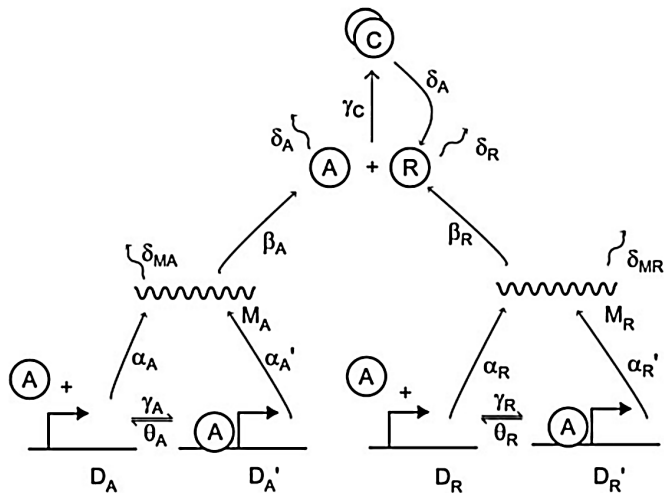
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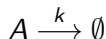
- ▶ VKBL model of circadian rhythms
- ▶ Deterministic and stochastic models
 - ▶ Comparable behaviour for $\delta_R = 0.2$
 - ▶ Qualitatively different behaviour for $\delta_R = 0.05$
- ▶ Explanation

VKBL model – parameters



$$\begin{aligned} \alpha_A &= 50 \text{ h}^{-1} \\ \alpha'_A &= 500 \text{ h}^{-1} \\ \alpha_R &= 0.01 \text{ h}^{-1} \\ \alpha'_R &= 50 \text{ h}^{-1} \\ \beta_A &= 50 \text{ h}^{-1} \\ \beta_R &= 5 \text{ h}^{-1} \\ \gamma_A &= 1 \text{ mol}^{-1} \text{ h}^{-1} \\ \gamma_R &= 1 \text{ mol}^{-1} \text{ h}^{-1} \\ \gamma_C &= 2 \text{ mol}^{-1} \text{ h}^{-1} \\ \delta_A &= 1 \text{ h}^{-1} \\ \delta_R &= 0.2 \text{ h}^{-1} \\ \delta_{MA} &= 10 \text{ h}^{-1} \\ \delta_{MR} &= 0.5 \text{ h}^{-1} \\ \theta_A &= 50 \text{ h}^{-1} \\ \theta_R &= 100 \text{ h}^{-1} \end{aligned}$$

[Vilar, Kueh, Barkai, Leibler, 2002]



$A(t)$... number of molecules in time t , provided $A(0) = n_0$

Law of mass action:

$$dA(t)/dt = -kA(t)$$

Gillespie stochastic simulation algorithm:

$kA(t)dt$... probability that a reaction occurs in $[t, t + dt)$

(a) $r \sim U(0, 1)$

(b) $\alpha = kA(t), \quad \tau = \frac{1}{\alpha} \ln \frac{1}{r}$

(c) $A(t + \tau) = A(t) - 1$

(d) $t := t + \tau$, go to (a)

$$dD_A/dt = \theta_A D'_A - \gamma_A D_A A$$

$$dD'_A/dt = -\theta_A D'_A + \gamma_A D_A A$$

$$dD_R/dt = \theta_R D'_R - \gamma_R D_R A$$

$$dD'_R/dt = -\theta_R D'_R + \gamma_R D_R A$$

$$dM_A/dt = \alpha'_A D'_A + \alpha_A D_A - \delta_{M_A} M_A$$

$$dM_R/dt = \alpha'_R D'_R + \alpha_R D_R - \delta_{M_R} M_R$$

$$dA/dt = \beta_A M_A + \theta_A D'_A + \theta_R D'_R \\ - A(\gamma_A D_A + \gamma_R D_R + \gamma_C R + \delta_A)$$

$$dR/dt = \beta_R M_R - \gamma_C A R + \delta_A C - \delta_R R$$

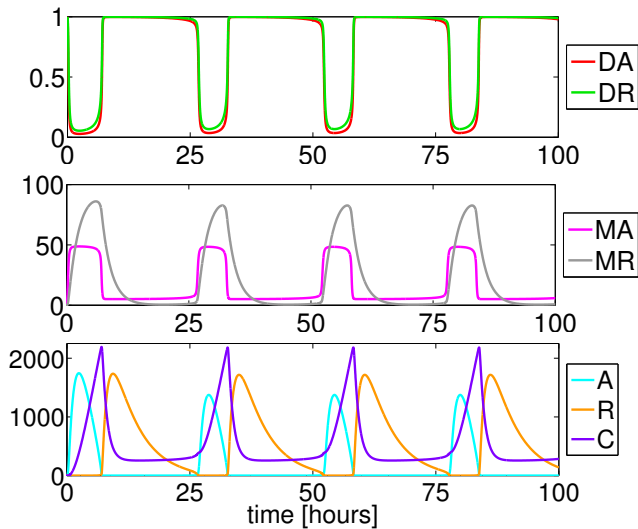
$$dC/dt = \gamma_C A R - \delta_A C$$

Initial conditions:

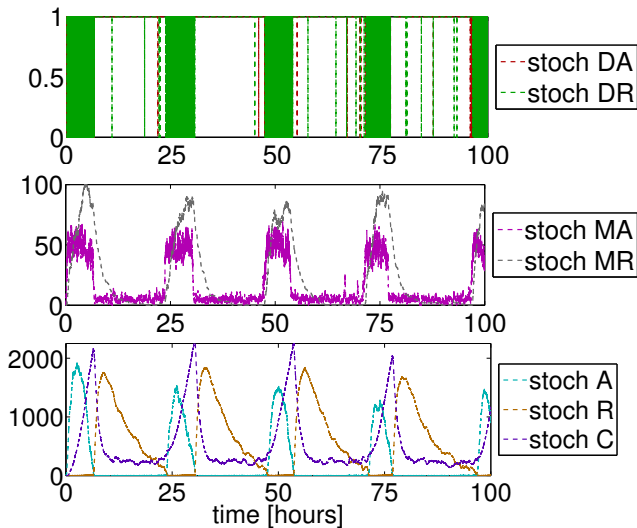
$$D_A = D_R = 1 \text{ mol}$$

$$D'_A = D'_R = M_A = M_R = A = R = C = 0 \text{ mol}$$

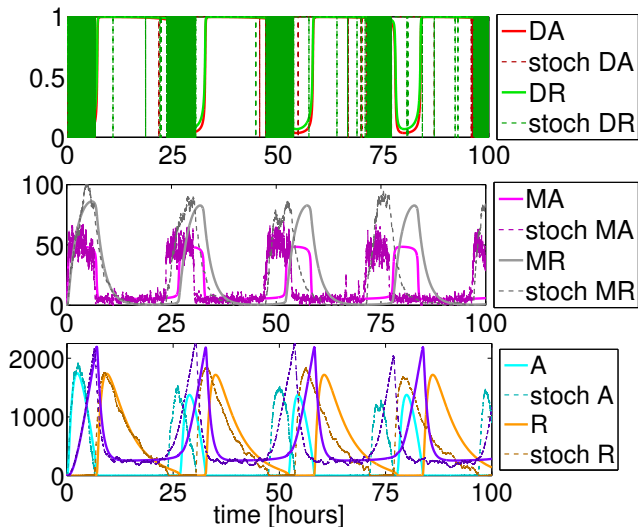
Full system – solution of ODE



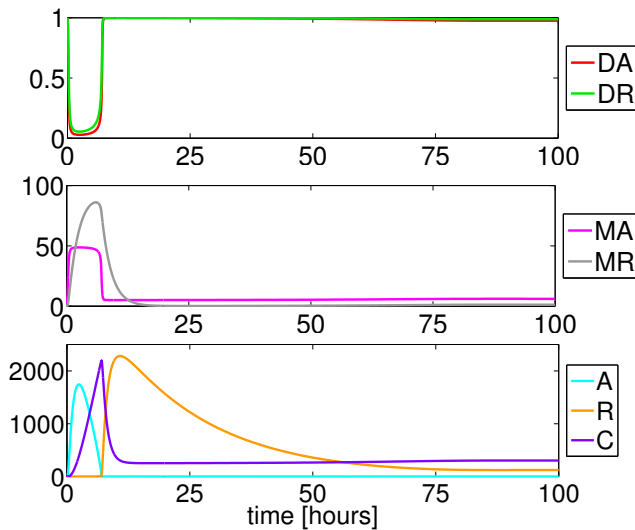
Full system – Gillespie SSA



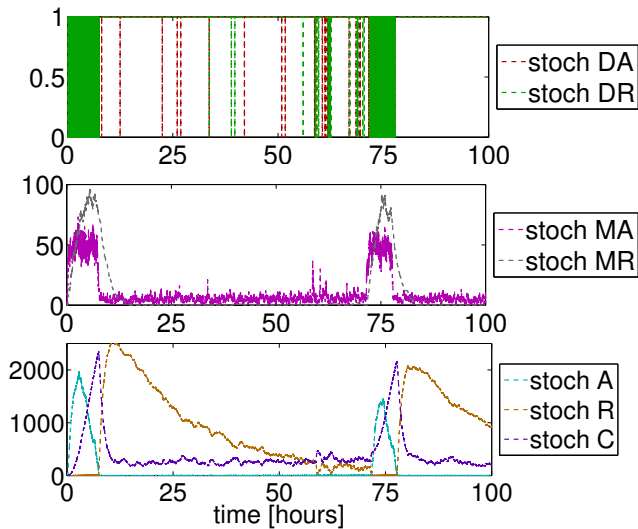
Full system – comparison



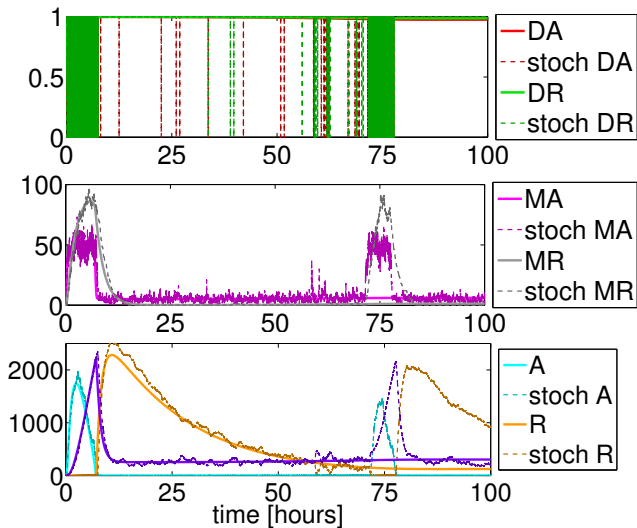
Full system – $\delta_R = 0.05$ – solution of ODE



Full system – $\delta_R = 0.05$ – Gillespie SSA

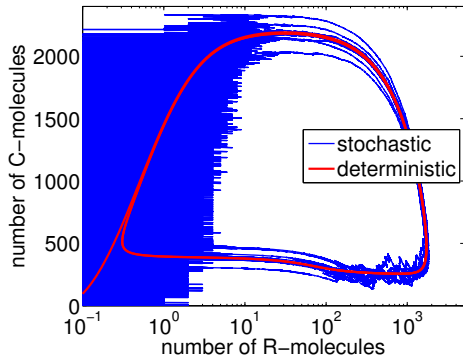


Full system – $\delta_R = 0.05$ – comparison

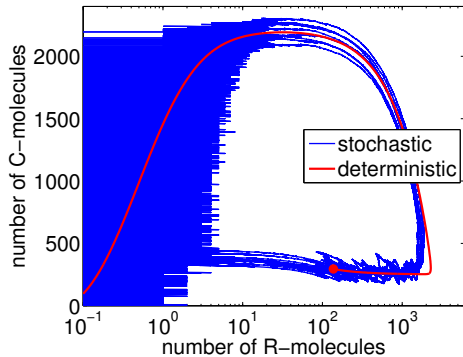


Full system – phase diagram

$\delta_R = 0.2$



$\delta_R = 0.05$



Quasi-steady state assumptions:

$$dR/dt = \beta_R \tilde{M}_R^s(R) - \gamma_C \tilde{A}^s(R)R + \delta_A C - \delta_R R$$

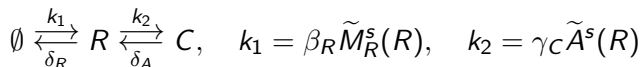
$$dC/dt = \gamma_C \tilde{A}^s(R)R - \delta_A C$$

$$\tilde{A}^s(R) = \frac{1}{2}(\alpha'_A \rho(R) - K_d) + \frac{1}{2}\sqrt{(\alpha'_A \rho(R) - K_d)^2 + 4\alpha_A \rho(R)K_d}$$

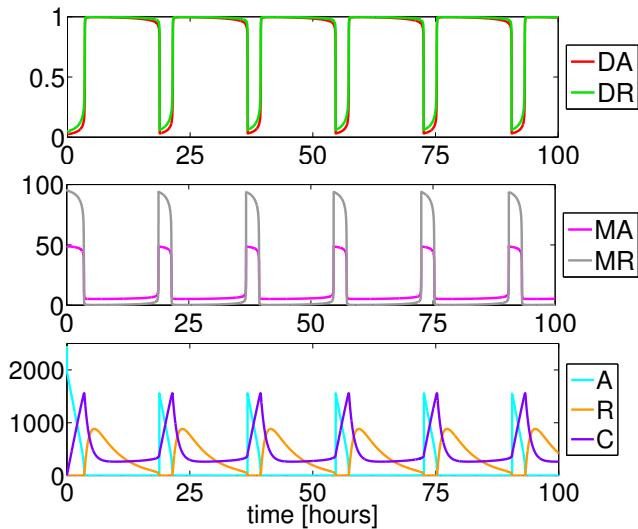
$$\rho(R) = \frac{\beta_A}{\delta_{M_A}} \frac{1}{\gamma_C R + \delta_A}, \quad K_d = \frac{\theta_A}{\gamma_A}$$

$$\tilde{M}_R^s(R) = \frac{\alpha'_R}{\delta_{M_R}} + \frac{\theta_R(\alpha_R - \alpha'_R)}{\delta_{M_R}(\theta_R + \gamma_R \tilde{A}^s(R))}$$

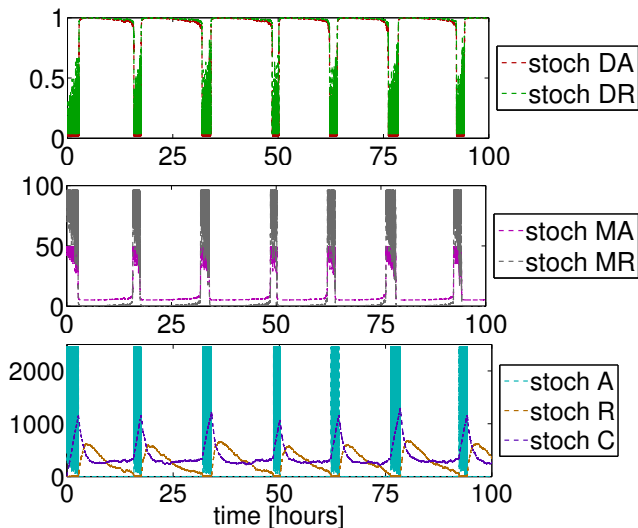
Simplified chemical system:



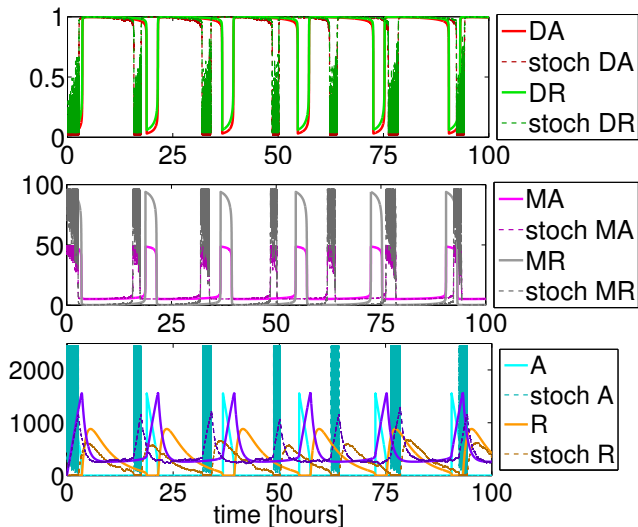
Simplified system – $\delta_R = 0.2$ – two ODE



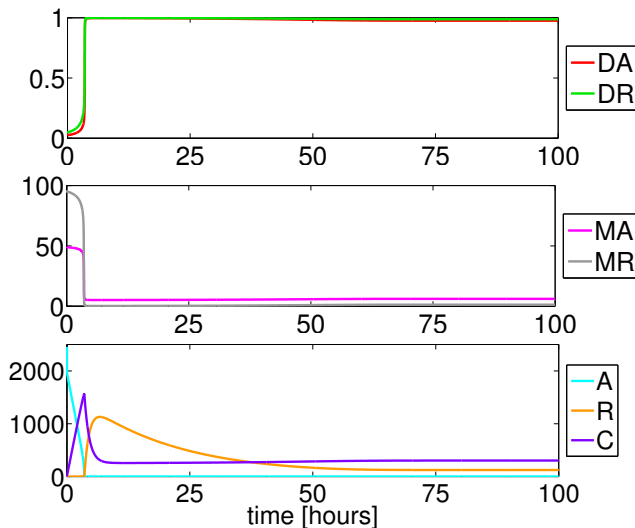
Simplified system – $\delta_R = 0.2$ – Gillespie SSA



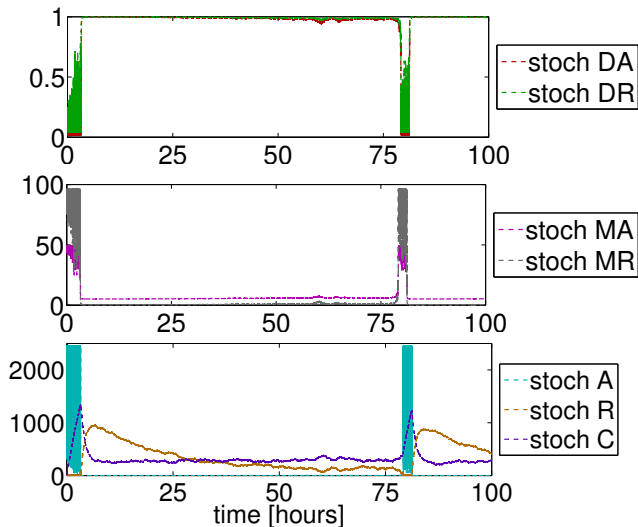
Simplified system – $\delta_R = 0.2$ – comparison



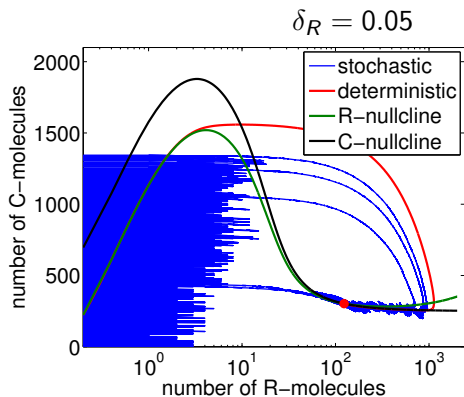
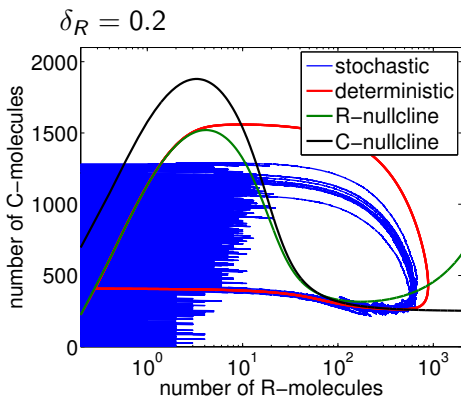
Simplified system – $\delta_R = 0.05$ – two ODE



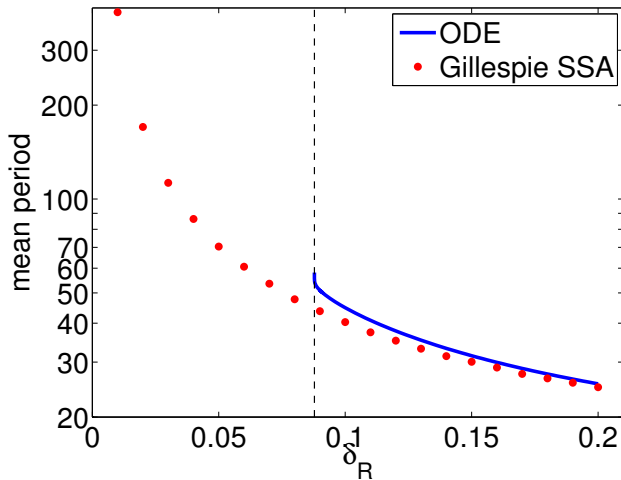
Simplified system – $\delta_R = 0.05$ – Gillespie SSA



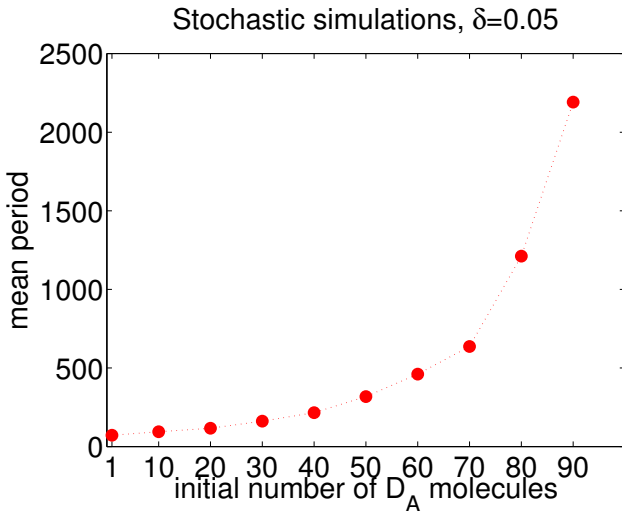
Simplified system – phase diagram



Mean period vs. δ_R



Mean period – $\delta_R = 0.05$



Period of deterministic system is infinite.

- ▶ Qualitatively different behaviour appears close to a bifurcation point
- ▶ Bifurcation point of stochastic system seems to be around $\delta_R = 0.001$
- ▶ Low copy numbers of a chemical species \implies stochastic effects
- ▶ High copy numbers \implies stochastic and deterministic models agree

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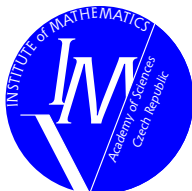
Thank you for your attention

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