

Nanotherapeutics Shielded With a pH Responsive Polymeric Layer

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Summary

Efficient intravenous delivery is the greatest single hurdle, with most nanotherapeutics frequently found to be unstable in the harsh conditions of the bloodstream. In the case of nanotherapeutics for gene delivery, viral vectors are often avidly recognized by both the innate and the adaptive immune systems. So, most modern delivery systems have benefited from being coated with hydrophilic polymers. Self-assembling delivery systems can achieve both steric and lateral stabilization following surface coating, endowing them with much improved systemic circulation properties and better access to disseminated targets; similarly, gene delivery viral vectors can be 'stealthed' and their physical properties modulated by surface coating. Polymers that start degrading under acidic conditions are increasingly investigated as a pathway to trigger the release of drugs or genes once the carrier reaches a slightly acidic tumor environment or after the carrier has been taken up by cells, resulting in the localization of the polymer in acidic endosomes and lysosomes. Advances in the design of acid-degradable drug and gene delivery systems have been focused and discussed in this article with stress placed on HPMA-based copolymers. We designed a system that is able to "throw away" the polymer coat after successful transport of the vector into a target cell. Initial biological studies were performed and it was demonstrated that this principle is applicable for real adenoviral vectors. It was shown that the transfection ability of coated virus at pH 7.4 is 75 times lower than transfection at pH 5.4.

Key words

Nanotherapeutics • Coating • pH responsive • Polymers • HPMA

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Introduction

Polymers are a popular choice for the design of drug delivery carriers. The use of polymer therapeutics has evolved into a broad discipline, employing a wide range of architectures either at the macroscale (gels and hydrogels) or nanoscale (polymer-drug conjugates, polymeric micelles, nanogels and cross-linked particles, polyplexes for DNA delivery, etc.). In particular, the development of multifunctional nanoscale devices has drawn a lot of attention. In summary, the controlled release of encapsulated therapeutics is an active research field. A few strategies have been developed, including bioconjugation with cell-targeting biomolecules for specific delivery and the use of stimuli-responsive polymers to create "smart" polymer therapeutics.

Quite a lot of stimuli-responsive nanotherapeutics have been designed for gene therapy. This may be because gene therapy provides new options for treating many human diseases in situations where conventional therapeutic approaches are not available (Rolland and Felgner 1998). However, major limitations that still restrict the therapeutic use of vectors include the quick elimination of the vectors from the bloodstream due to nonspecific interaction with plasma proteins, cells of the immune system and the uptake with cells of the

reticuloendothelial system (Lechardeur *et al.* 2005). Surface modifications (coating) to viral or nonviral vectors with covalently linked poly(ethylene glycols) (PEGs) and other hydrophilic polymers, which significantly prolong their circulation in plasma, ablate non-specific cell entry, evade contact with neutralizing antibodies, prevent binding to blood components and facilitate the transport of vectors to target cells, is a promising way to systemically deliver gene vectors (Dash *et al.* 2000, Eto *et al.* 2008, Fisher *et al.* 2001, Kreppel and Kochanek 2008, Oupický *et al.* 2002a, 2002b). Hydrophilic multivalent and semitelechelic *N*-(2-hydroxypropyl) methacrylamide copolymers (PHPMA) with 4-nitrophenoxy (ONp) and acyl thiazolidine-2-thione (TT) reactive groups enabling surface coating of both viral and nonviral gene delivery vectors have been developed and tested (Dash *et al.* 2000, Fisher *et al.* 2007, Fisher and Seymour 2010, Newman *et al.* 2009, Oupický *et al.* 2000, Sims *et al.* 2009, Šubr *et al.* 2006).

The multivalent reactive copolymer of HPMA and *N*-methacryloylglycylglycine 4-nitrophenyl ester was reacted with cationic liposome-DNA complexes (lipoplexes) to form an external polymer coat. Polymer coating increased the lipoplex diameter that was proportional to the amount of polymer used and saw a reduction in surface charge from strongly positive to neutral. Polymer-coated lipoplexes exhibited increased stability to protein binding compared to untreated liposomes and reduced non-specific uptake into in vitro cells (Papanicolaou *et al.* 2004).

Hydrophilic multivalent PHPMA copolymers with maleimide and pyridyl disulfide reactive groups were used to effect surface modification of adenovirus with a genetically introduced cysteine residue in the hyper-variable-region 5 of the hexon. A cysteine thiol group allows both irreversible attachment of shielding moieties via thioether bonds and reversible attachment via disulfide bonds (Prill *et al.* 2014).

Although covalently bound hydrophilic polymers sufficiently protect the vectors against undesirable interactions with blood-plasma components and efficiently mediate cellular uptake, the inability of these systems to throw away the shielding polymer layer after cell internalization results in a dramatic decrease in their transfection efficiency (Walker *et al.* 2005).

Therefore, new types of polymers for coating of gene delivery vectors have been developed. These polymers enable the stable coating of vectors during circulation in the blood stream, sufficient protection

against interactions with blood-plasma components and after cell internalization of polymer coated vectors, they enable the throwing away of the shielding polymer layer, thus restoring transfection efficiency (see [Scheme 1](#)).

The strategy of removing the shielding polymer layer from the surface of gene vectors is based on the hydrolysis of acid-labile bonds in endosomes (Meyer and Wagner 2006) or induced by an intracellular reductive environment (Oupicky and Diwadkar 2003, Oupicky *et al.* 2002a, 2002b, Saito *et al.* 2003).

Among the various existing stimuli-responsive systems, pH-sensitive polymers have been widely studied for drug and gene delivery applications. Indeed, it has been shown that the environment in tumor tissue is often 0.5-1.0 pH units lower than in normal tissue (Tannock and Rotin 1989). Moreover, a larger pH shift from 7.2-7.4 in the blood or extracellular spaces to 4.0-6.5 in the various intracellular compartments takes place during cellular uptake, and this can be used for intracellular drug or gene delivery.

Pegylation reagents containing acetal or ketal pH-sensitive shielding of DNA polyplexes were described by Knorr *et al.* (2007, 2008a, 2008b).

The bifunctional, endosomal pH-cleavable reagent OPSS-PEG-HZN-Chol [ω -2-pyridylidithio poly(ethylene) glycol α -(butyraldehyde) (N_1 -cholesteryl-oxycarbonyl-1,2-diaminoethaneamidocarboxy) pyridyl hydrazone] as a bioreversible surface shielding DNA lipopolyplexes was described by Nie *et al.* (2011).

Hydrophilic multivalent and semitelechelic PHPMA with TT reactive groups in their side chains or end of polymer chain bound to the polymeric backbone via disulfide spacers were used for studying the coating-decoating of amino-group modified polystyrene (PS) latex beads, PLL/DNA complexes (Carlisle *et al.* 2004, Laga *et al.* 2009), cowpea mosaic viruses (Laga *et al.* 2010), siRNA (Kostka *et al.* 2011) and adenoviruses (Carlisle *et al.* 2008). It was shown that incorporating a low content of quaternary ammonium groups into the TT reactive HPMA copolymers improved the coating efficiency of zwitterionic vesicles (Koňák *et al.* 2008) and adenoviruses (Carlisle *et al.* 2008, 2009, Šubr *et al.* 2009).

Methods

Chemicals

Methacryloyl chloride, 1-aminopropan-2-ol, 4,5-dihydrothiazole-2-thiol, 2-aminoethyl-trimethyl-

ammonium chloride hydrochloride, 4-oxopentanoic acid (OPA), 6-oxoheptanoic acid(OHA), 6-hydrazino-nicotinic acid (HNA), *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbo-diimide hydrochloride (DMAEC.HCl), *N,N'*-dicyclo-hexyl-arbodiimide (DCC), 4-(dimethylamino) pyridine (DMAP), triethylamine (Et₃N), trifluoracetic acid (TFA), 2,2'-azobis(isobutyronitrile) (AIBN), dichloromethane (DCM), tetrahydrofuran (THF), *N,N*-dimethylformamide (DMF), dimethyl sulfoxide (DMSO)and calcium hydride were from Sigma-Aldrich. 6-(*N*'-tert-butyloxycarbonyl-hydrazino)-nicotinic acid (Boc-HNA) was purchased from Merck. *N*-(3-Aminopropyl)methacrylamide hydrochloride was from Polysciences, Inc. All other chemicals and solvents were of analytical grade.

Monomer synthesis

N-(2-hydroxypropyl)methacrylamide (HPMA) was synthesized by modified reaction of methacryloyl chloride with 1-aminopropan-2-ol in dichloromethane in the presence of sodium carbonate as described by Ulbrich *et al.* (2000).

6-Methacrylamidohexanohydrazide (Ma-ah-NHNH₂) was prepared by a two-step procedure. Methyl 6-aminohexanoate hydrochloride was reacted with methacryloyl chloride in dichloromethane in the presence of sodium carbonate to form methyl-6-methacrylamidohexanoate. The Ma-ah-NHNH₂ was obtained by hydrazinolysis of methyl-6-methacrylamidohexanoate with hydrazine hydrate in methanol as described by Etrych *et al.* (2008).

2-Methyl-*N*-{5-[1-methyl-6-oxo-6-(2-thioxo-thiazolidin-3-yl)-hexylidene-hydrazinocarbonyl]-pentyl}-acrylamide (Ma-ah-NHN=OHA-TT) (see *Scheme 2*) was prepared by a two-step procedure as described in follows: Ma-ah-NHNH₂ (0.15 g, 0.703 mmol) was dissolved in methanol dried with CaH₂ (1 ml) and 6-oxoheptanoic acid (OheA) dried over P₂O₅ was added (0.113 g, 0.780 mmol) in the presence of 4-(1,1,3,3-tetramethylbutyl) pyrocatechol as inhibitor. Reaction was catalyzed by addition of acetic acid (5 ul) and carried out for 2 h at room temperature. The methanol was evaporated and the oily residue was dried for 1 h on vacuum. The oily residue was dissolved in dichloromethane (2 ml), 4,5-dihydrothiazole-2-thiol (0.093 g, 0.780 mmol) was added followed by addition of *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbodiimide hydrochloride (DMAEC· HCl) (0.180 g, 0.94 mmol) and catalytic amount of DMAP. Reaction mixture was stirred

for 3 h at room temperature. Reaction mixture was diluted with dichloromethane (10 ml) and extracted 3 times with NaHCO₃ solution (2 wt%, 10 ml). The organic layer was separated and dried with a Na₂SO₄. The dichloromethane was evaporated in vacuum the oily residue was dissolved in dried methanol (2 ml) and purified on Sephadex LH-20 column in methanol. The methanol was evaporated and oily product was dried on vacuum and stored in refrigerator. Yield 0.085 g (29 %). Elemental analysis: calc./found C=54.52/54.33 %; H=7.32/7.21 %; N=12.72/12.46 %; S=14.55/13.84 %.

1-(2-thioxo-thiazolidin-3-yl)-pentane-1,4-dione (OPA-TT) was prepared by reaction of 4-oxopentanoic acid (OPA) with 4,5-dihydrothiazole-2-thiol in the presence of *N,N'*-dicyclohexylcarbodiimide (DCC). 4-Oxopentanoic acid (2.85 g, 24.5 mmol) and 4,5-dihydrothiazole-2-thiol (2.93 g, 24.5 mmol) were dissolved in THF (20 ml) and cooled to -10 °C. *N,N'*-dicyclohexylcarbodiimide (DCC) (5.57 g, 27 mmol) was dissolved in THF (50 ml) and cooled to -10 °C. Cooled solutions were mixed together, catalytic amount of DMAP was added and reaction mixture was kept at -10 °C for 2 h and then at 5 °C for 16 h. Precipitated dicyclohexylurea was filtered off, THF was evaporated and product was crystallized from ethyl acetate. Yield: 3.5 g (72.0 %). M.p. 82 °C. Elemental analysis: calc./found C=44.22/44.34 %; H=5.1/5.23 %; N=6.45/6.47 %; S=29.51/28.87 %.

(6-hydrazino-pyridin-3-yl)-(2-thioxo-thiazolidin-3-yl)methanone hydrochloride (HNA-TT.HCl) was prepared in two steps according to following procedure: To a mixture of Boc-HNA (100.0 mg, 0.40 mmol) and TT (51.8 mg, 0.43 mmol) in ethyl acetate (15 ml) cooled on 4 °C was added DCC (51.8 mg, 0.43 mmol) dissolved in 5 ml ethyl acetate. The reaction was catalyzed by addition of a few crystals of DMAP and resulting mixture was kept at -18 °C for 2 h and then at 4 °C overnight. Afterwards, 0.1 ml of acetic acid was added and reaction mixture was stirred for 1 h at room temperature. The formed dicyclo-hexylurea was filtered off and the crude compound was purified by silica gel column chromatography using a gradient of EtOAc/hexan mixture as the eluent. The solvent was evaporated in vacuum, obtained residue was dissolved in 10 ml 4 M HCl/dioxan and stirred 3 h at room temperature. The resulting product was isolated by precipitation into Et₂O. Yield was 100.8 mg. Elemental analysis: calc./found C=37.17/36.28 %; H=3.81/3.42 %; N=19.27/19.11 %; S=22.05/21.86 %. ¹H-NMR (DMSO): δ = 3.27 (t, H2),

4.64 (t, H2), 7.20 (d, H1); 8.37 (d, H1); 8.92 (s, H1).

Polymer synthesis

The multivalent reactive copolymer poly(HPMA-co-Ma-Ah-NHN=OHA-TT) Cop 1 (see [Scheme 3](#)) was prepared by solution radical copolymerization of the HPMA (0.25 g, 1.75 mmol) and Ma-Ah-NHN=OHA-TT (0.085 g, 0.19 mmol) in DMSO (1.87 ml) initiated with AIBN (0.034 g) at 60 °C for 6 h (Šubr and Ulbrich 2006). Polymer was isolated by precipitation into mixture acetone diethylether (2:1), filtered off washed with acetone, diethylether and dried in vacuum. Yield 0.265 g. Content of TT groups 7.75 mol %, Mw 45 000 and Mw/Mn 2.36.

Synthesis of reactive multivalent terpolymer poly[(HPMA)-co-(AEMA-OPA-HNA-TT)-co(TMAEM)] Cop 5 (see [Scheme 4](#)) started from polymeric precursor poly[(HPMA)-co-(AEMA-BOC)-co-(TMAEM)], which was prepared by radical solution polymerization of HPMA (1.0 g, 6.98 mmol), BOC-AEMA (0.171 g, 0.79 mmol) and TMAEM (0.025 g, 0.12 mmol) in DMSO (5.83 ml) using ABIN (0.159 g, 0.97 mmol) as initiator. The polymerization was carried out at 60 °C for 16 h in a glass pressure tube under argon atmosphere. The polymer was isolated by precipitation into 20-fold excess of a diethyl ether/acetone (1:3) mixture, filtered off, dissolved in methanol and purified by Sephadex™ LH-20 size-exclusion chromatography in methanol. The purified polymer was precipitated with 20-fold excess of a diethyl ether, filtered off and incubated in the TFA : TIPS : H₂O (95:2.5:2.5) mixture (10 ml) for 2 h. The starting precursor was obtained by precipitation of reaction mixture into 20-fold excess of diethyl ether. Yield 1.054 g (88.1 %).

A mixture of 0.3 g poly[(HPMA)-co-(AEMA)-co-(TMAEM)] (0.19 mmol ~ NH₃⁺ groups) and 54.0 µl EDIPA (0.39 mmol) was dissolved in 2.0 ml DMSO and added drop-wise to a stirred solution of 50.7 mg OPA-TT (0.23 mmol) in 2.0 ml DMSO. After 12 min the reaction was stopped and the modified polymer was isolated by precipitation into a 40-fold excess of diethyl ether/acetone (1:1) mixture. The crude polymer was filtered off, dissolved in methanol and chromatographically purified on Sephadex™ LH-20 in methanol. The purified polymer was isolated by precipitation into a 40-fold excess of diethyl ether. The yield was 0.232 g (77.3 %).

To a solution of 0.15 g poly[(HPMA)-co-(AEMA-OPA)-co-(TMAEM)] (0.1 mmol ~(C=O)

groups) in 3.0 ml dimethylacetamid was drop-wise added HNA-TT.HCl dissolved in 1.0 ml dimethylacetamid. The reaction was catalyzed by addition of 1.0 µl concentrated H₂SO₄ and the mixture was stirred 24 h at room temperature. The resulting polymer was obtained by precipitation of the reaction mixture into a 40-fold excess of diethyl ether/acetone (1:3), filtered off, dissolved in methanol and chromatographically purified on Sephadex™ LH-20 in methanol. The purified polymer was isolated by precipitation into a 40-fold excess of diethyl ether. The yield was 0.139 g (92.7 %). Content of TT groups 6.0 mol. %, content of QA groups 1.2 mol. %, Mw 44 000, Mw/Mn 2.17.

The content of TT reactive groups in HPMA copolymers was determined spectrophotometrically in methanol ($\epsilon_{305}=10\ 800\ 1\text{ mol}^{-1}\text{ cm}^{-1}$). The weight-average molecular weight M_w and polydispersity M_w/M_n of copolymers were determined by SEC on HPLC Shimadzu with Superose 6™ column equipped with UV, differential refractometer Optilab® rEX and multiangle light scattering detector DAWN® 8™ (Wyatt Technology Corp., USA). 0.3 M sodium acetate buffer (pH 6.5) containing 0.5 g/l sodium azide was used as the mobile phase. The flow rate was 0.5 ml/min.

Coating of latex particles

0.5 ml (0.301 mg/ml) of the latex bead suspension and 20 ml of polymer solution containing the required amount of a coating polymer (freshly prepared in 10x10⁻³ M HEPES, pH=7.4) were mixed and then 20 ml of 1 M HEPES, pH=8.7 was added in order to adjust the pH to 8.2. The final concentrations of coating polymers in solution ranged from 0.062 to 4 mg/ml. The solution was stirred thoroughly, incubated for 30 min, diluted 1000 times with HEPES, pH=8.2, and finally 0.2 ml of the solution was transferred onto the chamber slide (Lab-Tek, Nunc, USA). Samples were measured immediately (within 1 h after the coating started). The solutions for DLS were prepared in the same way but in higher amounts (2 ml of latex suspension) and used without any dilution.

Dynamic and static light scattering measurements

Static light scattering (SLS) measurements were carried out with a homemade instrument equipped with a 30-mW He-Ne laser in the angular range 30-140°. The obtained data were analyzed using the Zimm plot procedure. Calculation of the amount of bound coating copolymer on the particle surface was made by the

procedure used in the previous paper. The accuracy of the Mw measurements was about 1 %. The refractive index increments of polystyrene latex particles (0.257 ml/g) and coating copolymers (0.167 ml/g) were taken from literature (Laga *et al.* 2007).

Polarized dynamic light scattering measurements were performed within the same instrument as mentioned above (SLS) with an ALV 5000 multibit autocorrelator. Data were analyzed using the GENDIST program. The experimental error of Rh-determination was typically about 3 %. Kinetics of coating processes was monitored through changes of the particle hydrodynamic radius (Rh) and scattering intensity (Is) at the scattering angle 173°

on a Nano-ZS, Model ZEN3600 (Malvern Instruments, UK) zetasizer.

In vitro experiments – monolayer infection

Studies were performed as described previously by Fisher *et al.* (2001). PC3 cells (96well plate, 104/well, 24 h) were infected with Ad5 or polymer coated Ad5 (500 copies/cell). After 90 min infection media were removed and washing in 150 µl PBS performed and fresh media added. 24 h later media were removed, washing (150 µl PBS) performed and luciferase measured (reporter system E151A (Promega) and a luminometer (LB9507, Berthold)).

Table 1. Characteristics of prepared PHPMA.

	Mw	Mw/Mn	TT	QA	Linker
Cop 1	37 500	1.9	5.5	0	OHA ^a
Cop 2	39 300	1.7	5.8	2.0	OHA
Cop 3	45 300	2.3	7.5	0	OHA
Cop 4	45 300	2.3	4.8	2.0	OHA
Cop 5	112 000	1.3	7.0	3.0	HNA ^b

^a 5-oxo-hexanoic acid, ^b hydrazino-nicotinic acid.

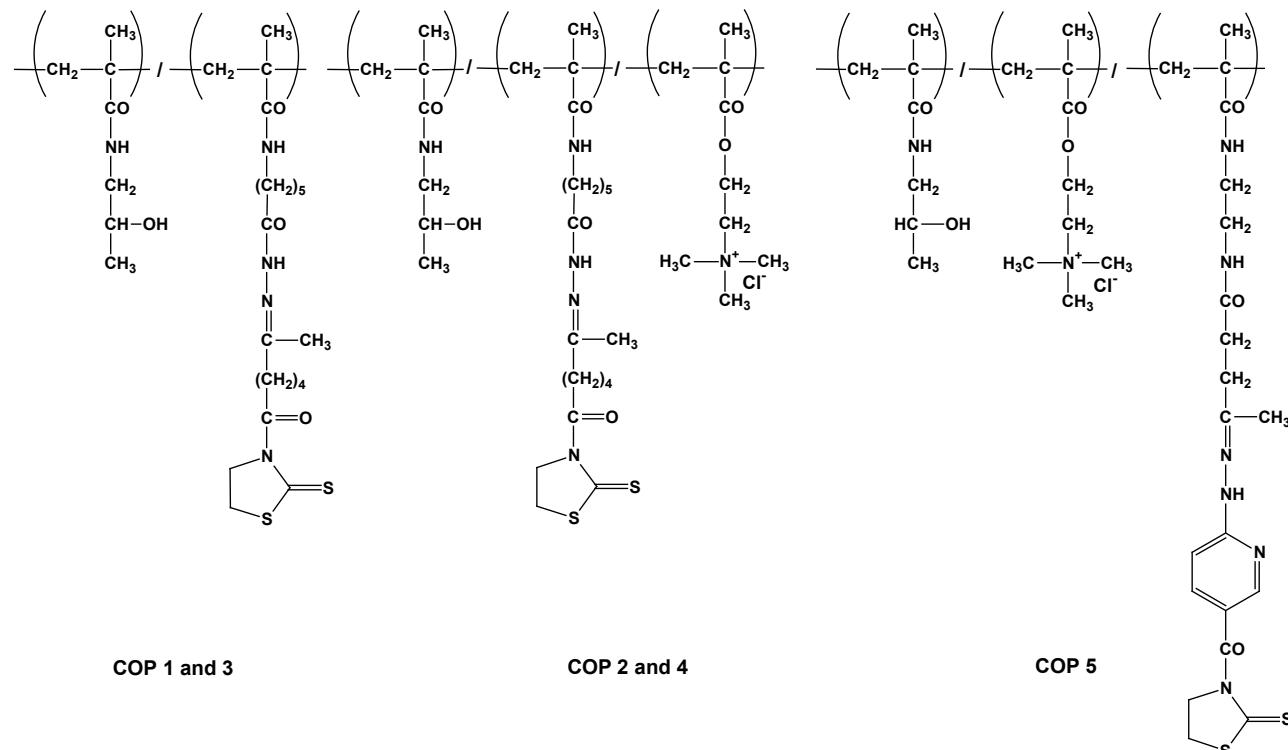


Fig. 1. Structures of synthetized HPMA-based copolymers.

Results and Discussion

In this paper, we describe the synthesis and physico-chemical characterization of HPMA-based copolymers containing reactive TT groups designed for surface modification of gene-delivery vectors. Multivalent PHPMA with TT groups randomly distributed along the polymer chain are employed and the TT groups are at the same time attached to the polymer backbone via spacers containing biodegradable pH-sensitive hydrazone bonds. We describe a detailed study of reactions enabling an efficient coating-decoating of gene delivery systems and a model system comprised of amino group-modified polystyrene latexes.

We synthesized appropriate monomers and novel PHPMA and we determined the potential of the copolymers in the synthesis of stimuli responsible nanotherapeutics. These new reactive copolymers were prepared by solution radical copolymerization of the respective TT-monomer with HPMA in DMSO using AIBN as an initiator. The physico-chemical characteristics of the polymers as molar mass, contents of reactive and quarterly ammonium (QA) groups are summarized in Table 1. The structures of the copolymers are depicted in Figure 1.

In principle, semitelechelic polymers allow for a well-defined coating of nanotherapeutics, especially polyelectrolyte complexes, as well as control of the quality of surface modification by varying the number of attached polymer chains and their length. However, multivalent copolymers enable a more efficient coating and better lateral stabilization of vectors by cross-linking the surface, although the structure of the surface layer is less defined. Here we would like to show how our polymers can modify the surface of Adenovirus 5 (Ad5).

Ad5 is one of the most commonly-used gene delivery vectors due to its highly efficient and regulated pathway of infection (Greber *et al.* 1993). However, its clinical application has been limited by the failure to gain full control of its tropism or ameliorate the strong innate and adaptive immune responses generated against it. One approach to taming Ad5 and improving its biocompatibility has been focused on making genetic modifications to its capsid proteins or to exchange them entirely for those of serotypes not normally associated with human infection (Roberts *et al.* 2006, Wickham *et al.* 1997). However, the cellular tropism of Ad5 is determined by a range of capsid proteins – cell receptor interactions, with an infection pathway that involves the knob and

shaft domains of the fiber (Bergelson *et al.* 1997). Similarly, the pre-existing immunity against Ad5 is both commonplace and targeted to a wide range of capsid epitopes (Sumida *et al.* 2005). Furthermore, fixation by complement and instigation of innate immunity may be achieved even in the absence of specific anti-adenovirus antibodies (Cichon *et al.* 2001). This raises the possibility that capsid proteins from ‘rare’ serotypes will still be inactivated by complement and IgM binding despite a lack of pre-exposure. Furthermore, it is the capsid as a whole that, following intravenous delivery, drives the strong interaction with the reticuloendothelial system (RES) and the resulting pro-inflammatory cytokine cascade, even in previously native mice and non-human primates (Muruve *et al.* 2004). Therefore, the use of genetic engineering to modify Ad5 and to create a platform delivery vector that is devoid of all mechanisms of natural infection, hidden from innate and pre-existing immune responses and does not trigger anti-capsid mediated adaptive immunity represents a substantial challenge. The complexity and magnitude of the genetic engineering necessary to meet these requirements necessitates the use of special producer cell lines and often leads to inefficiencies in production and the ultimate loss of infective capacity.

Chemical modification using reactive polymers to ‘stealth’ Ad5 offers a comparatively simple, efficient and non-heritable alternative to genetic modification that can be performed on virus post-production and purification.

The concept was first described by O'Riordan *et al.* (1999), who utilized a PEG bearing a single amine reactive group to demonstrate protection against neutralization and the maintenance of infection activity. Shortly after, Fisher *et al.* (2001) used a multivalent reactive polymer based on HPMA to enable the coating and retargeting of Ad5. This 16.5 kDa polymer presented multiple amine reactive ONp groups along its main chain, which was supposed to react with the capsid in a cooperative manner, enhancing coating without cross-linking neighboring Ad5 particles. It was found through ¹⁴C-trace labelling that more than 900 polymers coated each virus and that each polymer was attached at 14 points. It is notable that despite no attempt being made to achieve site specific capsid modification and little regard being paid to the influence of polymers on the intracellular translocation pathway of Ad, infective capacity was retained or redeemable in these studies. Before using of our new copolymer for Ad5 coating, we

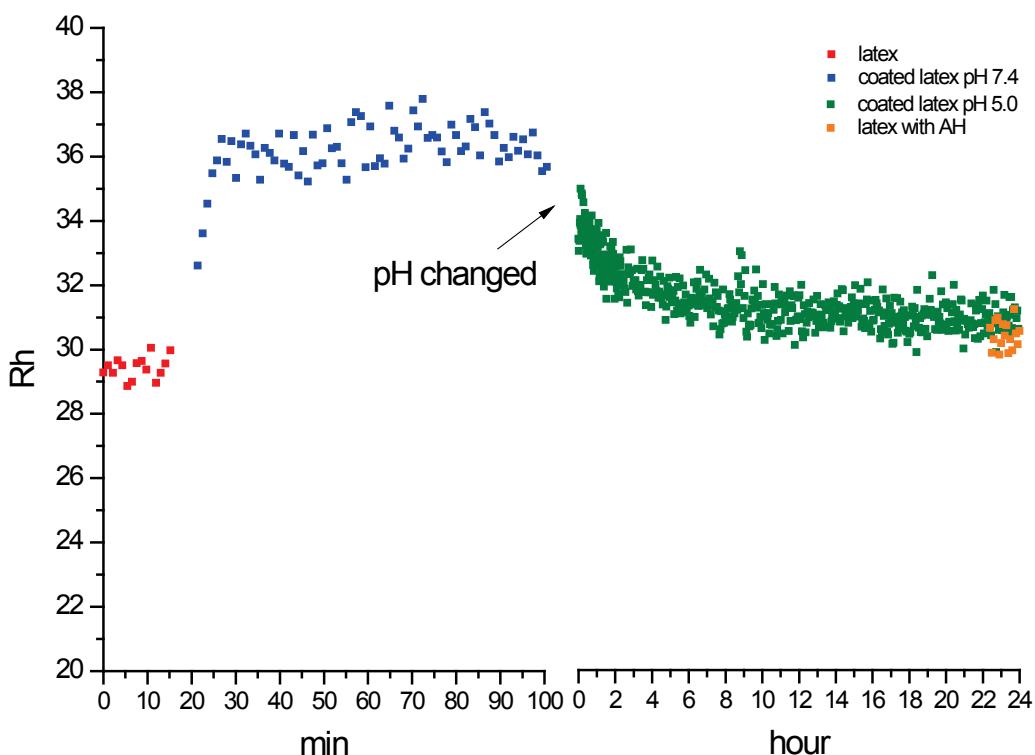


Fig. 2. The addition of Cop 1 to amino latex particles results in the surface modification of these particles, which is demonstrated by an increased hydrodynamic radius. Changes to pH evoke removal of the polymer coat from the latex particles, which results in the value of the hydrodynamic radius returning to the value of an unmodified latex. The reaction was performed in 10 mM HEPES pH 7.4. Data were obtained from DLS.

decided to prove the concept of a pH-responsive shielding layer on amino modified latex nanoparticles. Coating model latex particles bearing amino groups on their surface using a covalent coupling with acylating TT reactive groups incorporated in highly hydrophilic multivalent PHPMA structures were successfully demonstrated by changes in nanoparticle size (Fig. 2). Within 10 min after mixing latex beads with the shielding polymer, the hydrodynamic radius increased from 29 to 37 nm, which exactly corresponded to the size of the shielding polymer ($R_h \sim 4$ nm). The change in pH from 7.4 to 5.0 caused the polymer coat to be removed from the latex particles, which resulted in the value of the hydrodynamic radius to return to the value of unmodified latex.

Šubr *et al.* (2009) published findings that the inclusion of side chains bearing a small amount of positively charged QA groups improves the electrostatic interaction between the polymers with negatively charged Ad5 hexon proteins. Based on these findings, we decided to also incorporate QA groups in our copolymers (see characteristics in Table 1 and structures of synthetized copolymers in Figure 1). An adenovirus encoding luciferase reporter gene was coated with a range of

concentrations of cop 3 or 4 and then incubated at pH 7.4 or 5.0 before being added to prostate cancer PC3 cells. The level of transfected luciferase was quantified 24 h later. We found increasing shielding ability of polymers as concentrations increased (Fig. 3). When the same doses of shielding copolymers are compared, one can see the contribution of QA groups to the coating efficacy (decreased transfection). At 2 mg/ml, the difference is clearly visible. Unfortunately, the presence of QA groups slows down the efficacy of the deshielding procedure. The time dependence can clearly be seen in Figure 4. We used the highest amount of shielding polymers (20 mg/ml) from the previous study. So the transfection ability of Ad5 at pH 7.4 was decreased 2.5 times. Transfection activity increase of 1.5 times was achieved after 24 h of incubation at pH 5.0. But at this concentration of shielding polymers, we never saw the full restoration of transfection activity compared to an unmodified virus.

The fact that the restoration of transfection activity of shielded Ad depended on many factors is also demonstrated in Figure 5. There you can see that 2 or 5 h of pre-incubation of shielded Ad at varying pH increase the efficacy of transfection. If pH is changed from 7.4 to

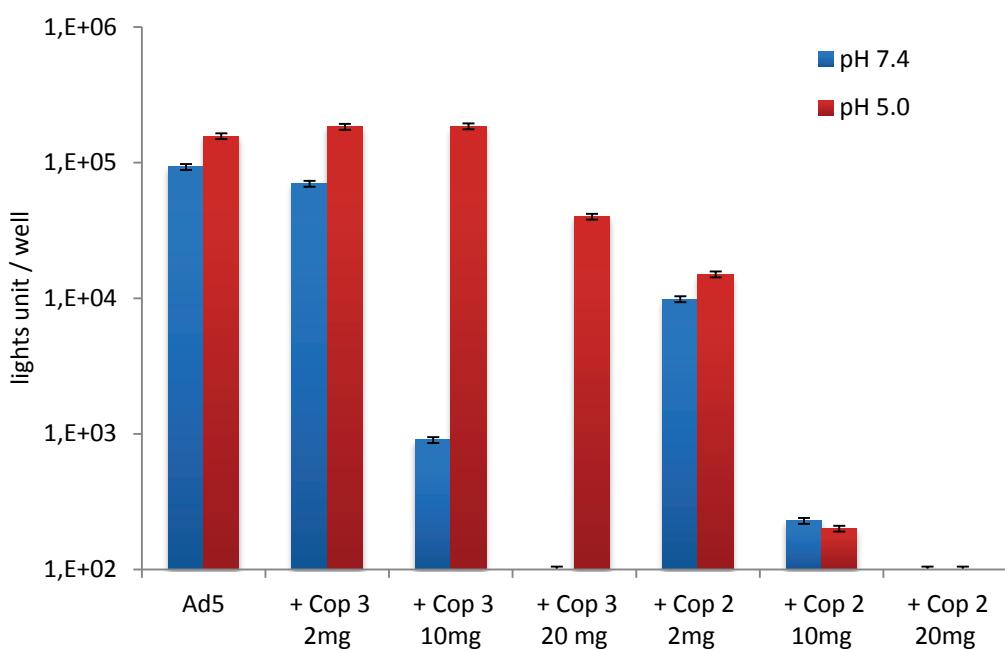


Fig. 3. An adenovirus encoding luciferase reporter gene was coated with a range of concentrations (10 or 20 mg/ml) of cop 2 or 3 and then incubated in PBS at pH 7.4 or 5.0 before being added to prostate cancer PC3 cells. Luciferase was quantified 24 hours later. N=3.

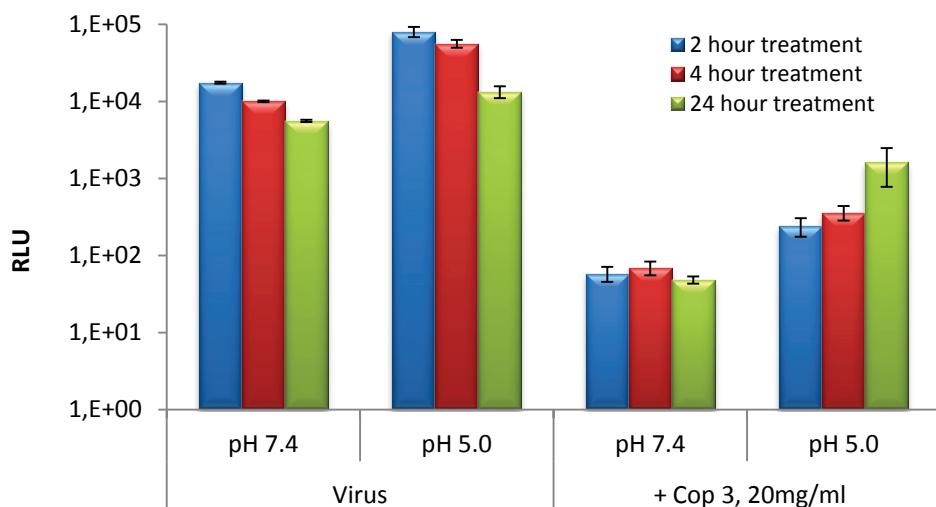


Fig. 4. An adenovirus encoding luciferase reporter gene was coated with copolymer 3 (20 mg/ml) and then incubated at in PBS at pH 7.4 or 5.0 before being added to prostate cancer PC3 cells for different time periods. Luciferase was quantified 24 hours later. N=3, SD shown.

5.4, the transfection efficacy is 2.5 orders of magnitude higher in the case of 2 h of pre-incubation and 2 orders of magnitude higher in the case of 5 h of preincubation. In this figure, you can see the varying stability of hydrazone bonds at different pH levels. The hydrazone bond shows the greatest amount of stability at pH 7.4 and is rapidly cleaved at pH 4.4. Hydrazino-nicotinic acid as a linker between the reactive group and the polymer backbone was used in this experiment.

Conclusions

We designed a system that is potentially able to “throw away” the polymer coat after successful transport of the vector into a target cell. Proof of this principle was shown in studies with model latex particles and confirmed with adenoviruses *in vitro* studies. The efficiency of coating nanoparticles with a reactive HPMA copolymer at pH 7.4 and decoating at pH 5.0 was determined by changes in the hydrodynamic radius (R_h) of uncoated and PHPMA-coated model nanoparticles by

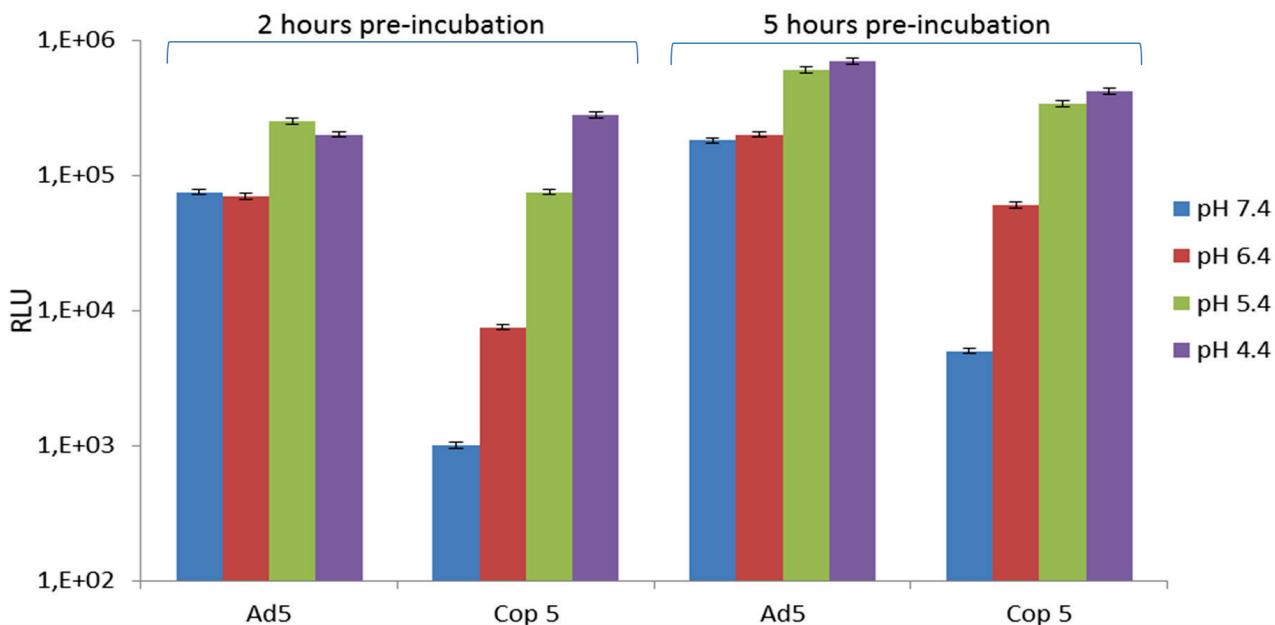


Fig. 5. Coating of Ad with Cop 5 in a scale of pH, after 2 or 5 hours pre-incubation in PBS with a corresponding pH. Data were obtained 24 hours post-incubation with PC3 cells.

static and dynamic light scattering methods. Initial biological studies were performed and it was demonstrated that this principle is applicable for real adenoviral vectors. Adenovirus type 5 was successfully coated with various concentrations of PHPMA with or without QA groups.

We demonstrated that after 2 h of incubating the coated virus (10 mg/ml Cop 3) with PC3 cells, the transfection ability of Ad5 at pH 7.4 was decreased 2.5 times compared to transfection at pH 5.0. In addition, the time dependence of the incubation of coated virus particles at different pH levels was proved. It was shown that the transfection ability of a coated virus at pH 7.4 is 75 times lower than transfection at pH 5.4 with 2 h of

pre-incubation at the corresponding pH. This can be considered evidence of pH responsible shielding.

Conflict of Interest

There is no conflict of interest.

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